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# Spectrally tunable near-infrared photoluminescence in $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$ ( $M = \text{Al, Ga, In}$ ) phosphate phosphors

Lipeng Jiang,<sup>a</sup> Xue Jiang,<sup>\*a,d</sup> Liangliang Zhang,<sup>id b</sup> Guocai Lv<sup>c</sup> and Yanjing Su<sup>id \*a</sup>

Modulation of the octahedral crystal field environment of  $\text{Cr}^{3+}$  is an effective approach to achieve tunable emission. Here, we prepared a series of broadband  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) near-infrared (NIR) phosphors, and cubic  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (APO-c: $\text{Cr}^{3+}$ ) and monoclinic  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (APO-m: $\text{Cr}^{3+}$ ) phosphors were prepared by controlling the synthesis temperature. The emission wavelength was tuned from 787 nm for APO-c: $\text{Cr}^{3+}$  to 894 nm for monoclinic  $\text{InP}_3\text{O}_9:\text{Cr}^{3+}$  (IPO: $\text{Cr}^{3+}$ ) by regulating the M ion and reducing the crystal field intensity. Excitingly, the  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) family shows excellent thermal stability; the emission intensity of APO-c: $\text{Cr}^{3+}$ , APO-m: $\text{Cr}^{3+}$  and monoclinic  $\text{GaP}_3\text{O}_9:\text{Cr}^{3+}$  (GPO: $\text{Cr}^{3+}$ ) can still maintain 95.6%, 86% and 86% of that at room temperature when heating to 423 K, respectively. An NIR LED device was prepared by incorporating GPO: $\text{Cr}^{3+}$  and a blue light LED, demonstrating the potential application in night vision and non-destructive testing.

## 1. Introduction

Near-infrared (NIR) spectroscopy is extensively used in night vision, health monitoring, non-destructive testing, and biological imaging due to its convenient and fast response characteristics.<sup>1–5</sup> The route of combining blue LEDs with NIR phosphors has been proven to be an effective and feasible approach, benefiting from the rich experience acquired during the development and application of white LEDs.<sup>6–8</sup> The phosphor performance almost ultimately determines the device performance; therefore, the development of broadband NIR phosphors that are thermally stable and can be efficiently excited by InGaN LED chips is crucial.<sup>9–11</sup>

The transition metal ion  $\text{Cr}^{3+}$  is considered as an ideal broadband NIR activator.<sup>12–15</sup> Firstly,  $\text{Cr}^{3+}$  has good absorption in the blue region and can be perfectly matched with blue chips. Additionally, the emission of  $\text{Cr}^{3+}$  depends on the octahedral crystal field, and a wide range of emission spectra can be tuned by modulating the octahedral crystal field strength.<sup>16–19</sup> Taking advantage of this property, researchers

have developed many  $\text{Cr}^{3+}$ -activated NIR-emitting materials, and these studies have focused on aluminates, borates, niobates/tantalates, *etc.*<sup>20–23</sup> For instance, in our previous study, we replaced the  $\text{Al}^{3+}\text{-Ga}^{3+}$  pair with  $\text{Mg}^{2+}\text{-Ge}^{4+}$  in  $\text{Gd}_3\text{Al}_2\text{Ga}_3\text{O}_{12}:\text{Cr}^{3+}$ , and the emission wavelength shifted from 729 nm to 805 nm.<sup>24</sup> Zhang *et al.* reported  $\text{ABO}_4:\text{Cr}^{3+}$  ( $A = \text{Ga, Sc, In; B} = \text{Ta, Nb}$ ) NIR phosphors, which achieved wide spectral tuning from NIR-I to NIR-II with the peak position from 825 nm to 1025 nm by cation replacement.<sup>25</sup> Zhong *et al.* achieved an emission peak redshift from 726 nm to 830 nm by replacing the  $\text{Ga}^{3+}\text{-Ga}^{3+}$  unit with  $\text{Mg}^{2+}\text{-Ge}^{4+}$  in the  $\text{Ga}_2\text{O}_3$  structure.<sup>26</sup>

Recently,  $\text{Cr}^{3+}$ -activated phosphates have attracted much attention due to the simple preparation method and the low price of raw materials.<sup>27–30</sup> The weak crystal field generated by their relatively relaxed crystal structure offers the possibility for achieving broadband NIR emission. Researchers have reported a variety of  $\text{Cr}^{3+}$ -doped phosphate NIR phosphors. For example, several researchers reported  $\text{AMP}_2\text{O}_7$  ( $A = \text{Li, Na, K; M} = \text{Al, Ga, Sc, In}$ ) phosphors, which can be tuned in the 780–900 nm range by varying the A and M ions.<sup>31–35</sup> Zhao *et al.* reported  $\text{Sr}_9\text{M}(\text{PO}_4)_7:\text{Cr}^{3+}$  ( $M = \text{Ga, Sc, In}$ ) phosphors, and investigated the effect of structural constraints on the luminescence performance.<sup>36</sup> Miao *et al.* reported tunable broadband NIR photoluminescence by co-doping  $\text{Ga}^{3+}$  in  $\text{Na}_3\text{Sc}_2(\text{PO}_4)_3:\text{Cr}^{3+}$  phosphors.<sup>37</sup> Unfortunately, almost all developed phosphate phosphors exhibit poor thermal stability.

In this paper, we report a series of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) NIR phosphate phosphors. The effect of the synthesis temperature on the  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  phase transition was investigated. A wide range of spectral tuning (787–894 nm) was achieved by

<sup>a</sup>Beijing Advanced Innovation Center for Materials Genome Engineering, Corrosion and Protection Center, University of Science and Technology Beijing, Beijing, 100083, China. E-mail: jiangxue@ustb.edu.cn, yjsu@ustb.edu.cn

<sup>b</sup>State Key Laboratory of Luminescence and Applications, Changchun Institute of Optics, Fine Mechanics and Physics, Chinese Academy of Sciences, 3888 Eastern South Lake Road, Changchun 130033, China

<sup>c</sup>Basic Experimental Center of Natural Science, University of Science and Technology Beijing, Beijing, 100083, China

<sup>d</sup>Collaborative Innovation Center of Steel Technology, University of Science and Technology Beijing, 100083 Beijing, China

varying the M ion. In addition, the  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga}$ ) family shows excellent thermal stability; the emission intensity of cubic  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (APO-c: $\text{Cr}^{3+}$ ), monoclinic  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (APO-m: $\text{Cr}^{3+}$ ) and monoclinic  $\text{GaP}_3\text{O}_9:\text{Cr}^{3+}$  (GPO: $\text{Cr}^{3+}$ ) can still maintain 95.6%, 86% and 86% of that at room temperature when heated to 423 K, respectively. NIR LED devices prepared using GPO: $\text{Cr}^{3+}$  for night vision and non-destructive testing confirmed that  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) are promising NIR phosphors.

## 2. Materials preparation and characterization

The  $\text{MP}_3\text{O}_9:0.04\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) phosphors were obtained by a solid state reaction method.  $\text{Al}_2\text{O}_3$  (99.99%),  $\text{Ga}_2\text{O}_3$  (99.99%),  $\text{In}_2\text{O}_3$  (99.99%),  $\text{NH}_4\text{H}_2\text{PO}_4$  (99%), and  $\text{Cr}_2\text{O}_3$  (99.99%) were taken as raw materials. All the raw materials were weighed according to the chemical formula and then mixed and ground for 20 min. The raw materials were held in a muffle furnace at 500 °C for 2 h, and then heated up again to 800–1000 °C for 5 h. After cooling down, the  $\text{MP}_3\text{O}_9:0.04\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) phosphors were obtained.

The NIR LED devices were fabricated by coating blue LED chips (450–460 nm) with a mixture of the prepared phosphors and silicone (A : B = 1 : 4). The devices were further cured at 100 °C for 1 h and 150 °C for 2 h in an oven.

The XRD curves were obtained using an X'PERT PRO MPD X-ray diffractometer. The emission spectra, excitation spectra, temperature dependent emission spectra and luminescence decay times were acquired *via* an FLS-1000 fluorescence spectrophotometer. The absolute PL QY measurement (Hamamatsu C9920-02) was applied to obtain the IQE, a 450 nm laser was used as excitation sources for  $\text{MP}_3\text{O}_9:0.04\text{Cr}^{3+}$  ( $M = \text{Al, Ga}$ ) and

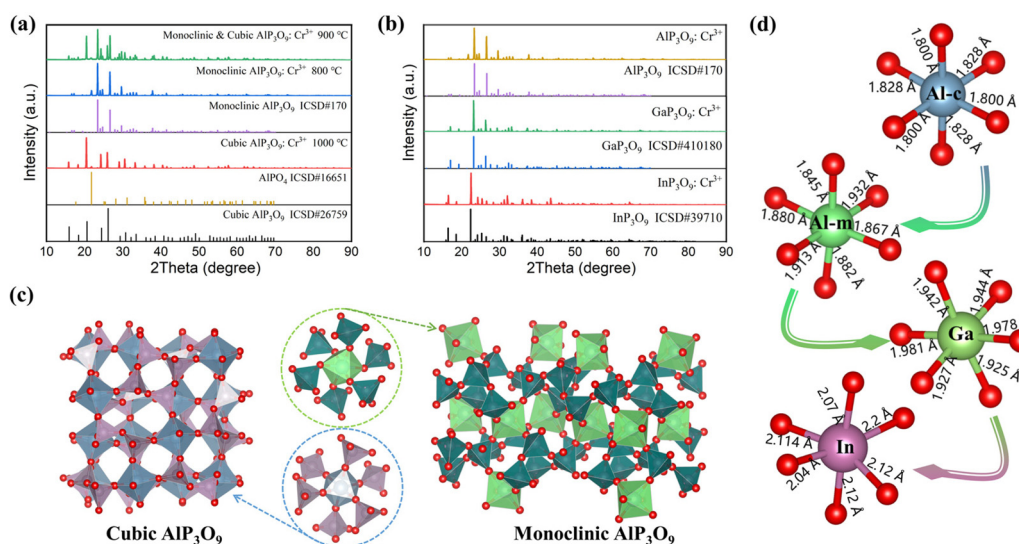
a xenon lamp was used as the excitation source for  $\text{InP}_3\text{O}_9:\text{Cr}^{3+}$ . The fabricated LED's output power was evaluated using an OHSP-350M (Hopoo Optoelectronics Co., Ltd, 350–1050 nm). Meanwhile, photographs were acquired using an industrial USB night vision camera (Shenzhen Zhongweiaoke Technology Co. Ltd, 1080 P, 700–1100 nm, China) to demonstrate the application of NIR pc-LEDs.

## 3. Results and discussion

### 3.1. Crystal structure of $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$

Fig. 1a displays the XRD patterns of  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  obtained at different sintering temperatures. By comparing with the standard card, it is found that the sample obtained at 900 °C contains both cubic  $\text{AlP}_3\text{O}_9$  (AIP-c) and monoclinic  $\text{AlP}_3\text{O}_9$  (AIP-m). The sample sintered at 1000 °C shows cubic phase, which corresponds to the standard card ICSD#26759, and the sample sintered at 800 °C shows monoclinic phase, which corresponds to the standard card ICSD#170. This indicates that the material system undergoes a phase transition from AIP-m to AIP-c with increasing temperature. A small amount of  $\text{AlPO}_4$  (ICSD#16651) phase appeared in the sample due to the decomposition of phosphate at high temperature. According to the domain knowledge,  $\text{Cr}^{3+}$  preferentially occupies the  $[\text{AlO}_6]$  octahedral site, while  $\text{AlPO}_4$  assumes the  $[\text{AlO}_4]$  tetrahedral structure; hence the presence of  $\text{AlPO}_4$  exerts no influence on the photoluminescence property.

The XRD patterns of monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ) phosphors are given in Fig. 1b. The diffraction peaks shift to a lower angle when the M ion changes from Al to In, which is attributed to the lattice expansion caused by the increase of the M ion radius. As shown in Fig. 1d, the average M–O bond length in the  $[\text{MO}_6]$  octahedron increases gradually as the M



**Fig. 1** (a) XRD patterns of cubic  $\text{AlP}_3\text{O}_9$  and monoclinic  $\text{AlP}_3\text{O}_9$ . (b) XRD patterns of monoclinic  $\text{MP}_3\text{O}_9:0.04\text{Cr}^{3+}$  ( $M = \text{Al, Ga, In}$ ). (c) Crystal structure and coordination environment of  $\text{AlO}_6$  octahedra of cubic  $\text{AlP}_3\text{O}_9$  and monoclinic  $\text{AlP}_3\text{O}_9$ . (d) The M–O bond length of  $\text{MP}_3\text{O}_9$  ( $M = \text{Al, Ga, In}$ ).  $\text{Al-cO}_6$  and  $\text{Al-mO}_6$  are octahedra in cubic and monoclinic  $\text{AlP}_3\text{O}_9$ , respectively.

ion changes from Al to In. The diffraction angle decreases when the crystal plane spacing  $d$  increases according to the Bragg equation  $2d\sin\theta = n\lambda$ . The crystal structure and coordination environment of  $[\text{AlO}_6]$  octahedra of cubic  $\text{AlP}_3\text{O}_9$  and monoclinic  $\text{AlP}_3\text{O}_9$  are given in Fig. 1c. In the  $\text{MP}_3\text{O}_9$  crystal structure, the  $[\text{MO}_6]$  octahedron is surrounded by six  $[\text{PO}_4]$  tetrahedra, forming a larger  $[\text{MP}_6]$  octahedron by sharing O ions with dot connections. The average M–O bond length of the  $[\text{MO}_6]$  octahedron increases gradually (Fig. 1d), which enables the change of the local crystal field environment and thus the tuning of the emission spectrum.

### 3.2. Octahedron-dependent tunable photoluminescence

Fig. 2a shows the emission and excitation spectra of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  (M = Al, Ga, In). The excitation spectra of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  phosphors exhibit two excitation bands at 400–500 nm and 600–800 nm, corresponding to the  ${}^4\text{A}_{2g} \rightarrow {}^4\text{T}_{1g}$  and  ${}^4\text{A}_{2g} \rightarrow {}^4\text{T}_{2g}$  transitions of  $\text{Cr}^{3+}$ , respectively. The emission spectra that cover 600–1300 nm originate from the  ${}^4\text{T}_{2g} \rightarrow {}^4\text{A}_{2g}$  transition of  $\text{Cr}^{3+}$ . With the sequential changes of M ions from Al to Ga and In, the emission spectrum is gradually red-shifted, and the emission peak position is red-shifted from 787 nm to 894 nm. The redshift of the emission spectrum is associated with a decrease in the intensity of the crystal field. The crystal field intensity can be calculated by eqn (1):<sup>38,39</sup>

$$D_q = \frac{1}{6} Z e^2 \frac{r^4}{R^5} \quad (1)$$

where  $D_q$ ,  $Z$ ,  $e$ ,  $r$  and  $R$  are the crystal field strength, anion charges, electron charges, the radius of the d-wave function and the Cr–O bond length, respectively. According to the analysis of Fig. 1d, the Cr–O bond length increases gradually from APO-c to IPO, leading to a decrease in  $D_q$ .

In general, it exhibits broadband emission when  $\text{Cr}^{3+}$  is in a weak crystal field environment. The strongness of the crystal field environment can be evaluated by  $D_q/B$ , where  $D_q$  is the crystal field strength and  $B$  is the Racah parameter. It is generally accepted that the criterion for a weak crystal field environment is  $D_q/B < 2.3$ ; eqn (2) can be used to calculate  $D_q/B$ .<sup>6,40</sup>

$$\begin{aligned} 10 \cdot D_q &= E_a({}^4\text{T}_{2g}) \\ \frac{D_q}{B} &= \frac{15 \cdot (x - 8)}{x^2 - 10 \cdot x} \\ D_q \cdot x &= E_a({}^4\text{T}_{1g}) - E_a({}^4\text{T}_{2g}) \end{aligned} \quad (2)$$

where  $E_a({}^4\text{T}_{1g})$  and  $E_a({}^4\text{T}_{2g})$  are the energy positions of  ${}^4\text{T}_{1g}$  and  ${}^4\text{T}_{2g}$ , respectively. In this study,  $E_a({}^4\text{T}_{2g})$  was calculated as the median of the peak energies of the  ${}^4\text{A}_{2g} \rightarrow {}^4\text{T}_{2g}$  excitation band and  ${}^4\text{T}_{2g} \rightarrow {}^4\text{A}_{2g}$  emission band, taking into account the correlation between the Stokes shift and bandwidth. The calculated  $D_q/B$  of APO-c: $\text{Cr}^{3+}$ , APO-m: $\text{Cr}^{3+}$ , GPO: $\text{Cr}^{3+}$ , and IPO: $\text{Cr}^{3+}$  are 1.66, 1.57, 1.51, and 1.44, respectively. Thus, the emission spectra of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  (M = Al, Ga, In) all show broadband emission. The detailed photoluminescence parameters can be found in Table 1.

The relationship between the emission intensity of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  (M = Al, Ga, In) phosphors and the M ion is depicted in Fig. 2b. The photoluminescence intensity decreases gradually when the M ion changes from Al to In. Meanwhile, the full widths at half maximum (FWHM) of their emission bands increase gradually. The FWHMs of APO-c: $\text{Cr}^{3+}$ , APO-m: $\text{Cr}^{3+}$ , GPO: $\text{Cr}^{3+}$ , and IPO: $\text{Cr}^{3+}$  are 120 nm, 139 nm, 141 nm, and 166 nm, respectively. The broadened FWHM indicates a strong electron–phonon coupling effect, and in this case, non-radiative decay depends on the Stokes shift  $\Delta R$ . As M changes from Al to In, the  $\Delta R$  gradually increases (Table 1), leading to an

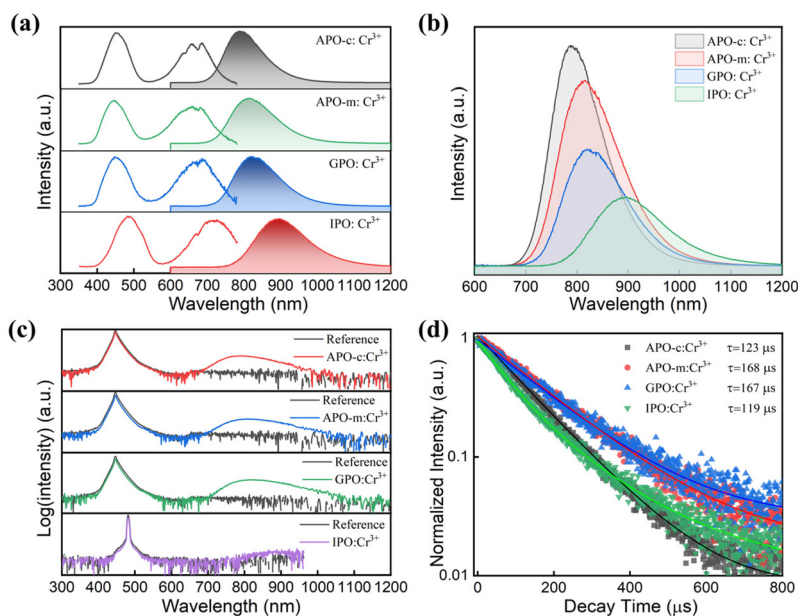


Fig. 2 (a) Normalized emission and excitation spectra, (b) comparison of the emission intensity, (c) IQE and (d) decay times of  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  (M = Al, Ga, In).

**Table 1** PL properties of Cr<sup>3+</sup>-doped MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In) NIR phosphors

Phosphors	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	$D_{\text{q}}$ (cm <sup>-1</sup> )	$B$ (cm <sup>-1</sup> )	$\Delta R$ (cm <sup>-1</sup> )	$D_{\text{q}}/B$
APO-c:Cr <sup>3+</sup>	450/670	787	1384	834	2263	1.66
APO-m:Cr <sup>3+</sup>	450/672	813	1359	866	2580	1.57
GPO:Cr <sup>3+</sup>	450/676	819	1350	892	2582	1.51
IPO:Cr <sup>3+</sup>	480/725	894	1249	867	2607	1.44

increase in the non-radiative transition probability, resulting in a decrease in the emission intensity.<sup>31</sup> Accordingly, the IQE of APO-c:Cr<sup>3+</sup>, APO-m:Cr<sup>3+</sup>, GPO:Cr<sup>3+</sup>, and IPO:Cr<sup>3+</sup> are 43.5%, 33.6%, 30.2% and 19.8%, respectively (Fig. 2c). Fig. 2d shows the decay curves of MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In) phosphors. All the decay curves can be fitted well by single exponential function, which illustrates that the NIR emission at room temperature stems from one luminescence center. The lifetime of APO-c:Cr<sup>3+</sup> is calculated to be 123  $\mu$ s, which is at the same level as previously reported.<sup>41</sup> The lifetimes of APO-m:Cr<sup>3+</sup>, GPO:Cr<sup>3+</sup>, and IPO:Cr<sup>3+</sup> are 168  $\mu$ s, 167  $\mu$ s, and 119  $\mu$ s, respectively. The lifetime of APO-c:Cr<sup>3+</sup> is shorter than that of monoclinic MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In), indicating that APO-c:Cr<sup>3+</sup> suffers from stronger nonradiative relaxation at room temperature.<sup>42</sup>

### 3.3. Temperature-dependent luminescence of MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup>

The thermal stability of phosphors is crucial for their applications, especially phosphate phosphors, which generally suffer from the challenge of poor thermal stability.<sup>43,44</sup> The emission spectra of MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In) phosphors at different temperatures are displayed in Fig. 3a–d. The emission intensity decreases with increasing temperature, which is due to thermal quenching. The integrated intensity of APO-c:Cr<sup>3+</sup> remains at 95.3% of that at room temperature when heating to

423 K, which is the highest value reported so far for Cr<sup>3+</sup>-activated phosphate materials such as KAlP<sub>2</sub>O<sub>7</sub>:Cr<sup>3+</sup> (77% @423 K),<sup>33</sup> KGaP<sub>2</sub>O<sub>7</sub>:Cr<sup>3+</sup> (56% @423 K),<sup>34</sup> and Na<sub>3</sub>Sc<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>:Cr<sup>3+</sup> (84% @423 K).<sup>37</sup> The integrated intensity of monoclinic MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In) phosphors at 423 K maintains 86%, 86%, and 38% of that at room temperature, respectively, which are also superior to those of most phosphate NIR phosphors (Fig. 3e).

The temperature dependent emission peak positions of both APO-c:Cr<sup>3+</sup> and APO-m:Cr<sup>3+</sup> phosphors exhibit a small red shift, which is attributed to the lattice expansion with increasing temperature. In addition, the FWHM of the emission spectrum changes significantly with increasing temperature, and the broadening of the FWHM at high temperature is associated with the electron–phonon coupling effect. The Huang–Rhys factor ( $S$ ) can be used to reflect the variation of electron–phonon coupling, which can be calculated by the following equation:<sup>45</sup>

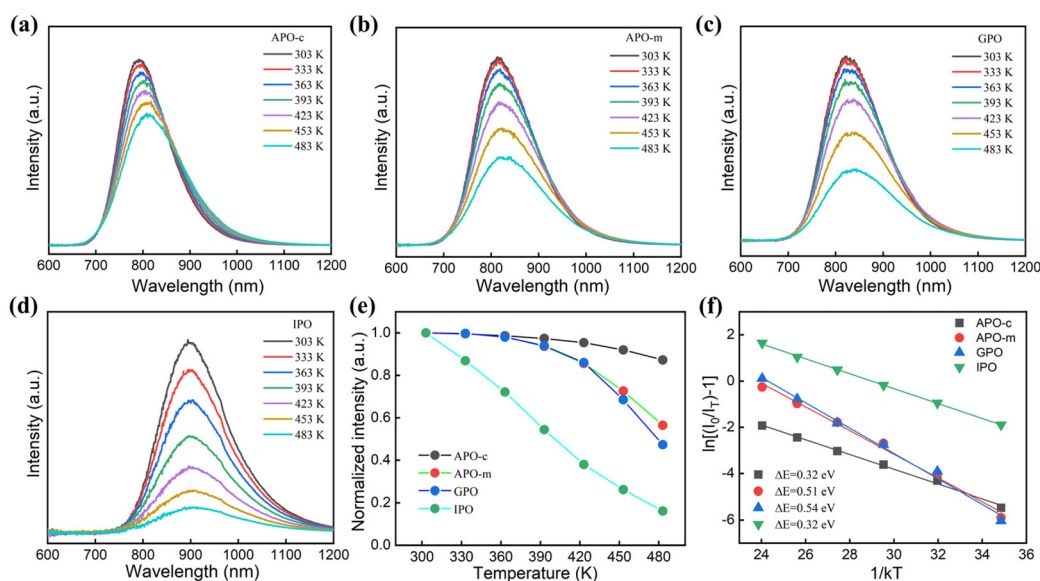
$$\text{FWHM} = 2.36\sqrt{S}\hbar\omega\sqrt{\cot h\left(\frac{\hbar\omega}{2kT}\right)} \quad (3)$$

where  $\hbar\omega$  and  $k$  are the effective phonon energy and Boltzmann constant, respectively. Eqn (3) reflects the variation of the FWHM with temperature, and it can be simplified as eqn (4).

$$\text{FWHM}^2 = 5.57 \times S \times (\hbar\omega)^2 \left[ 1 + \frac{1}{\hbar\omega/2kT} \right] \quad (4)$$

Then, if we let  $a = 5.57 \times S \times (\hbar\omega)^2$  and  $b = 5.57 \times S \times \hbar\omega$ , eqn (4) can be further simplified as eqn (5).

$$\text{FWHM}^2 = a + \frac{b}{1/(2kT)} \quad (5)$$



**Fig. 3** The emission spectra of (a) APO-c:Cr<sup>3+</sup>, (b) APO-m:Cr<sup>3+</sup>, (c) GPO:Cr<sup>3+</sup>, and (d) IPO:Cr<sup>3+</sup> at 303–483 K. (e) Normalized emission intensity at different temperatures and (f) the calculated activation energy of MP<sub>3</sub>O<sub>9</sub>:Cr<sup>3+</sup> (M = Al, Ga, In).



The Huang–Rhys factors of monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ) phosphors are calculated to be 4.5, 4.67, and 8.25, respectively, and 8.63 for cubic  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (Fig. 4). Generally, the stronger the electron–phonon coupling effect, the larger the lattice relaxation, and the thermal quenching activation energy ( $\Delta E$ ) is smaller. The activation energy ( $\Delta E$ ) can be obtained by the Arrhenius equation (eqn 6).<sup>46–48</sup>

$$I_T = \frac{I_0}{1 + C \exp\left(-\frac{\Delta E}{kT}\right)} \quad (6)$$

where  $I_T$  denotes the emission intensity of the sample at temperature  $T$ , and  $I_0$ ,  $k$  and  $C$  are the initial emission intensity, Boltzmann constant and constant, respectively. The activation energies of monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ) phosphors are 0.51 eV, 0.54 eV and 0.32 eV, respectively, and 0.32 eV for  $\text{APO-c}:\text{Cr}^{3+}$  (Fig. 3f). These values are higher than those of previously reported phosphate phosphors such as  $\text{KAlP}_2\text{O}_7:\text{Cr}^{3+}$  (0.292 eV),<sup>33</sup>  $\text{AlP}_3\text{O}_9:\text{Cr}^{3+}$  (0.29 eV),<sup>41</sup>  $\text{KGaP}_2\text{O}_7:\text{Cr}^{3+}$  (0.3 eV),<sup>34</sup> and  $\text{LiGe}_2(\text{PO}_4)_3:\text{Cr}^{3+}$ .<sup>27</sup> The decrease in activation energy when  $\text{M}$  changes from  $\text{Al}$  to  $\text{In}$  is attributed to the large Stokes shift that reduces the energy difference between the lowest point of the excited state and the intersection of ground and excited states. The decrease in activation energy will reduce the energy barrier for non-radiative transitions, leading to a decrease in thermal stability. The changes of the activation energy and Huang–Rhys factor in monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ) phosphors are observed to coincide with the trend of thermal stability. Hence, the enhanced electron–phonon coupling effect and the reduced energy barrier contribute together to the reduced thermal stability of monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ) phosphors when  $\text{M}$  changes from  $\text{Al}$  to  $\text{In}$ .

$\text{APO-c}:\text{Cr}^{3+}$  exhibits excellent thermal stability; however, it suffers from a larger electron–phonon coupling effect and a lower activation energy compared with  $\text{APO-m}:\text{Cr}^{3+}$ . This paradox may suggest that the electron–phonon coupling associated with the nonradiative transition plays a non-dominant role in the thermal quenching of  $\text{APO-c}:\text{Cr}^{3+}$ . In a recent

work, Huang *et al.* attributed the excellent thermal stability of  $\text{APO-c}:\text{Cr}^{3+}$  phosphors to negligible thermal expansion, which is well supported by their experimental data.<sup>41</sup> In fact, the thermal stability of the phosphor exhibited is the result of the combined effect of the band gap, electron–phonon coupling, structural rigidity, *etc.* As we analyzed in Fig. 1c, the  $[\text{MO}_6]$  octahedron is surrounded by six  $[\text{PO}_4]$  tetrahedra in the  $\text{MP}_3\text{O}_9$  crystal, forming a larger  $[\text{MP}_6]$  octahedron. This unique structure may give  $[\text{MO}_6]$  great rigidity, which may explain its excellent thermal stability. As shown in Fig. 1d,  $\text{APO-c}:\text{Cr}^{3+}$  possesses a more compact octahedral environment compared to monoclinic  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ), suggesting a rigid structure. The minimum Stokes shift also confirms this, as the Stokes shift is strongly affected by the rigidity of the crystal structure. Further research is needed on the thermal quenching mechanism of the  $\text{MPO-c}:\text{Cr}^{3+}$  family. To summarize,  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $\text{M} = \text{Al}, \text{Ga}, \text{In}$ ) phosphors all show good thermal stability and can be considered as candidate materials for NIR LEDs.

#### 3.4. Photoluminescence of $\text{GaP}_3\text{O}_9:x\text{Cr}^{3+}$

Comprehensively considering the luminescence intensity, thermal stability and wavelength, the  $\text{GaP}_3\text{O}_9:\text{Cr}^{3+}$  phosphor was selected for a detailed investigation. To optimize the  $\text{Cr}^{3+}$ -doping concentration,  $\text{GPO}:x\text{Cr}^{3+}$  ( $x = 0.01\text{--}0.1$ ) were prepared. The XRD patterns of  $\text{GPO}:x\text{Cr}^{3+}$  are presented in Fig. 5a. All the XRD patterns agree with the standard card, proving the successful realization of the pure phase. With increasing doping concentration, the emission intensity increases first and then decreases, and  $x = 0.06$  was identified as the optimum doping concentration (Fig. 5b). The decrease in intensity at a high doping concentration is attributed to concentration quenching. In addition, the lifetime gradually decreases from 168.2  $\mu\text{s}$  to 163  $\mu\text{s}$  with increasing  $\text{Cr}^{3+}$  concentration, suggesting that the possibility of nonradiative transitions increases with increasing  $\text{Cr}^{3+}$  concentration (Fig. 5c).

The temperature dependent emission spectra of the optimal doping concentration sample are given in Fig. 5d.

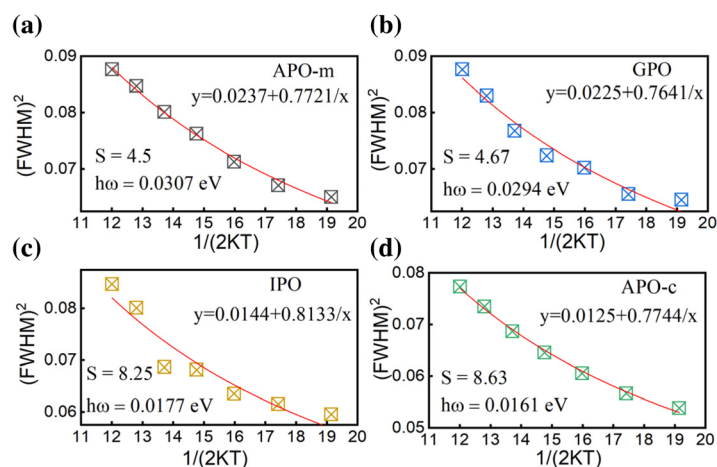
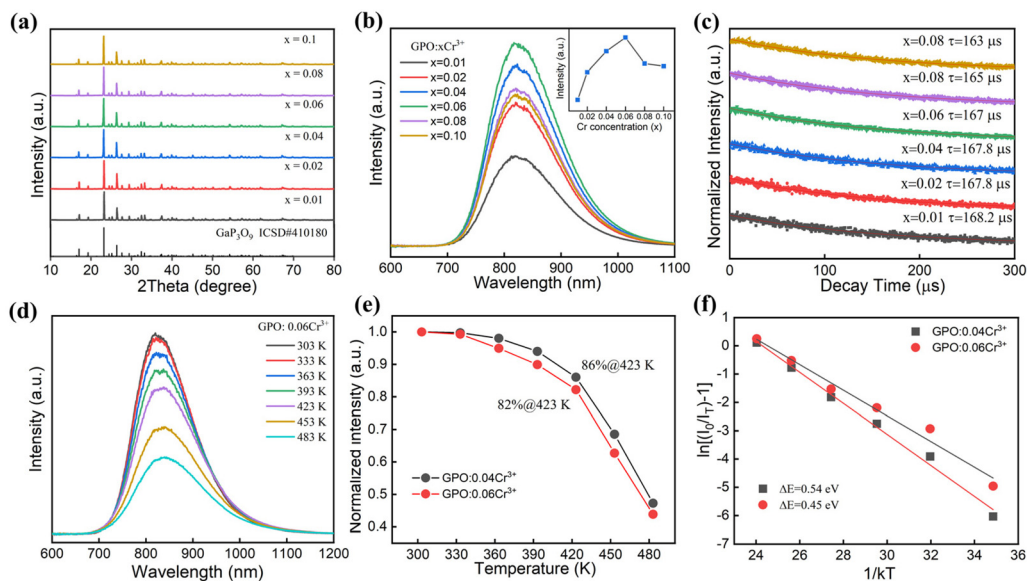


Fig. 4 Fitting  $(\text{FWHM})^2$  and  $1/(2KT)$  to obtain the Huang–Rhys factor. (a)  $\text{APO-m}:\text{Cr}^{3+}$ , (b)  $\text{GPO}:\text{Cr}^{3+}$ , (c)  $\text{IPO}:\text{Cr}^{3+}$ , and (d)  $\text{APO-c}:\text{Cr}^{3+}$ .



**Fig. 5** (a) XRD patterns, (b) emission spectra and (c) decay times of GPO: $x\text{Cr}^{3+}$ . (d) Temperature-dependent emission spectra of GPO:0.06 $\text{Cr}^{3+}$ . (e) Normalized emission intensity of GPO: $x\text{Cr}^{3+}$  ( $x = 0.04, 0.06$ ) at different temperatures. (f) Activation energy of GPO: $x\text{Cr}^{3+}$  ( $x = 0.04, 0.06$ ).

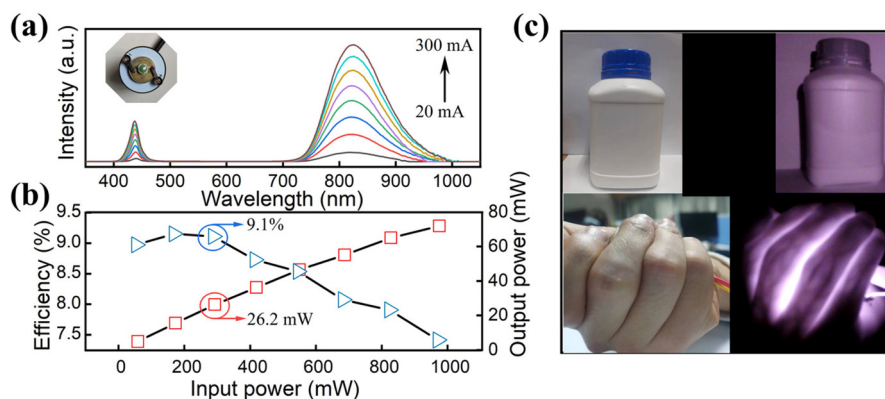
With increasing temperature, the emission peak position is redshifted from 820 to 843 nm and the FWHM is broadened from 139 to 165 nm due to electron–phonon coupling. In addition, Fig. 5e shows the thermal stability of GPO: $x\text{Cr}^{3+}$  versus the doping concentration. The results show that GPO:0.04 $\text{Cr}^{3+}$  possesses better thermal stability as the probability of energy transfer to the quenching center is low at a low concentration. The activation energies show the same pattern, with the activation energies calculated to be 0.54 and 0.45 eV for GPO:0.04 $\text{Cr}^{3+}$  and GPO:0.06 $\text{Cr}^{3+}$  (Fig. 5f). The large activation energies provide energy barriers against non-radiative transitions and ensure excellent thermal stability.

### 3.5. NIR pc-LED devices and application

To demonstrate the application of our prepared NIR phosphors in the NIR region, GPO: $\text{Cr}^{3+}$  was chosen to be combined

with 450 nm blue LEDs to prepare NIR LED devices. The emission spectra of the NIR LED device are shown in Fig. 6a. The inset shows the fabricated LED device. The emission spectra contain a narrow band emission of 450 nm blue light and a broadband emission of GPO: $\text{Cr}^{3+}$  centered at 820 nm. The emission intensity increases with the increase of the input power, while the photoelectric conversion efficiency increases first and then decreases gradually. The device produces an output power of 26.2 mW when the input power is 288 mW, and the photoelectric conversion efficiency is 9.1% (Fig. 6b).

In addition, we also show the application of the prepared pc-LED in night vision and non-destructive testing, as shown in Fig. 6c. In natural light, a visible camera can take a colored photo of the chemical bottle, but it is not possible to observe the remaining amount of chemicals, and there is no image when it is dark. In contrast, the NIR camera captures a black



**Fig. 6** (a) Electroluminescence spectra of the NIR-pc-LED made using GPO: $\text{Cr}^{3+}$  at different input currents. The inset shows the fabricated LED device. (b) The output power and photoelectric conversion efficiency as a function of input power. (c) Application demonstration of the NIR light source.

and white image when illuminated with NIR LEDs. Additionally, benefiting from the excellent penetration, the NIR light source can be used to check the remaining amount of chemicals. NIR light can also easily penetrate the fingers and we can capture the blood vessels with an NIR camera due to the difference in NIR light absorption by human tissues. All these results show that the NIR light source has great potential in night vision and non-destructive testing.

## 4 Conclusion

In conclusion, a series of broadband  $\text{MP}_3\text{O}_9:\text{Cr}^{3+}$  ( $M = \text{Al}, \text{Ga}, \text{In}$ ) NIR phosphors were successfully synthesized. The phase transition of  $\text{AlP}_3\text{O}_9$  can be controlled by controlling the sintering temperature, with APO-m formed at 800 °C and APO-c formed at 1000 °C. With the variation of M ions from Al to In, the crystal field intensity decreases gradually and the emission wavelength redshifts from 787 nm to 894 nm. APO-m: $\text{Cr}^{3+}$  and GPO: $\text{Cr}^{3+}$  exhibit good thermal stability (86%@423 K) owing to their high activation energy and unique rigid structure. APO-c: $\text{Cr}^{3+}$  shows excellent thermal stability (95.3%@423 K), and its thermal quenching mechanism still needs further investigation. The NIR device prepared using GPO: $\text{Cr}^{3+}$  produces an output power of 26.2 mW when the input power is 288 mW, and the photoelectric conversion efficiency is 9.1%, demonstrating the potential application in night vision and non-destructive testing. Our work provides more promising possibilities to further develop phosphate phosphors with excellent thermal stability.

## Conflicts of interest

There are no conflicts to declare.

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