

Optimized impedance matching and enhanced attenuation by heteroatoms doping of yolk-shell CoFe₂O₄@HCN as highly efficient microwave absorbers

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ABSTRACT

Microwave absorbing (MA) materials with yolk-shell structures have been extensively studied in impedance matching. However, the impedance matching achieved by the complementary effect of the core and the shell does not determine the reflection of the microwaves upon the occurrence at the first incidence. The interaction between the outer layer of materials and the electromagnetic waves significantly impacts the MA properties of materials. In this study, the impedance matching improvement method of the shell structure was further explored by preparing CoFe₂O₄@HCN (honeycomb carbon with N-doping) through the hydrothermal method followed by hydrolysis, polymerization, etching, and annealing. The resulting structure with heteroatoms doping provided the novel CoFe₂O₄@HCN with excellent impedance matching and multiple loss mechanisms contributing to MA process. The absorber with a filler loading of 40% exhibited an RL_{min} of −68.03 dB with a matching thickness of 2.5 mm. The efficient absorbing bandwidth reached 5.92 GHz (a change from 11.92 to 17.84 GHz) at 1.99 mm thickness. Interestingly, these findings look promising for future synthesis and application of yolk-shell structure microwave absorbers.

1. Introduction

Over the years, the development of 5G communication technology has gradually advanced along with the 6G technique, resulting in serious electromagnetic pollution [1–3]. Flying electromagnetic waves are dangerous to human health and life and interfere with the operation of electronic equipment [4,5]. To reduce the influence of electromagnetic waves, microwave absorbing (MA) materials have been widely studied in various fields as effective materials, which are divided into conduction loss type, dielectric loss type, and magnetic loss type [6]. For applications, a preeminent MA material should simultaneously satisfy “strong absorbing capacity, wide efficient absorbing bandwidth (EAB), thin matching thickness, and low density” [7,8].

Carbon has been extensively investigated and considered as a research concern as MA material due to its good stability, low density, and diverse morphological characteristics. Examples include zero-dimensional (0D) carbon quantum dots [9], 1D carbon nanotubes [10,

11], 2D graphene [12–14], 3D carbon spheres [15], and carbon composite materials [16,17]. In particular, carbon spheres and nanocubes possess the best plasticity and mesoporous morphology, which is conducive to impedance mismatching and reflection of electromagnetic waves. Zhang et al. [18] fabricated and applied hollow carbon microspheres with adjustable mesopores. The unique structure provided scattering and multi-reflection inside the absorbers, leading to favorable MA performance with the minimum reflection loss (RL_{min}) of −30.4 dB at thickness of 3.6 mm. Chen et al. [19] fabricated Fe@NCNs (N-doped carbon nanocubes)/MnO₂ yolk-double shell materials and investigated the heterogeneous interface resulting from phase separation and the mechanism of MA properties of the absorbers. The results showed that the RL_{min} of −46.7 dB was achieved at 3.3 mm, and the EAB reached 10.8 GHz at 3.1 mm thickness. Magnetic materials are considered as one of the most widely studied MA materials. The excellent magnetism provides magnetic loss, and magnetic materials can be combined with dielectric and conductive materials to achieve synergistic effect

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[20–22]. As a spinel-type transition metal oxide, CoFe_2O_4 exhibits superior magnetic loss performance with certain dielectric loss, thereby it is widely used in the synthesis of MA materials. For example, Cheng et al. [23] applied a biomimetic layered absorber composed of carbon and CoFe_2O_4 with RL_{\min} value reaching -53.54 dB at 2.46 mm, and maximum EAB reached 4.92 GHz at 1.7 mm. The excellent absorbing performances are attributed to the combined action of dielectric loss, conductive loss, and magnetic loss, resulting in the synergistic effect of hollow structure and interface effect. MA materials with a single structure and component generally possess poor impedance matching, narrow absorbing bandwidth, and low reflection loss. In general, MA materials with the yolk-shell structure are effective for optimizing the impedance matching and increasing the multiple reflections responsible for absorbing electromagnetic waves. Moreover, the interfacial relaxation of yolk-shell structure resulting from hetero-interfaces-induced polarization has been proven to be conducive to enhancing the absorbing performance [24]. Examples of such materials include $\text{Fe}/\text{m-SiO}_2/\text{PPy}$ [25], $\text{Co}@/\text{SiO}_2@\text{C}$ [26], $\text{air}@/\text{Co}@/\text{Co}_3\text{Sn}_2@\text{SnO}_2$ [27], and $\text{C}@/\text{void}@/\text{C}$ [28]. However, despite the extensive research efforts devoted to the study on the yolk-shell materials, most research has been focused on the overall impedance matching through internal and external synergism; nonetheless, the effect of impedance matching between the outer shell and free space has not been thoroughly investigated. The doping of nonmetallic atoms in carbon has been shown to improve impedance matching. For instance, Jiang et al. [29] reported S-NCFs (S/N atoms-doped carbon nanofibers) with massive number of dipoles to improve the dipole polarization effect of the absorber and thus optimized its impedance matching performance. The resulting S-NCFs exhibited the RL_{\min} value of as low as -48.67 dB at 16.1 GHz, while the thickness was 1.39 mm with a mass ratio of 15%. The doping of heteroatoms based on carbon helps in designing particular yolk-shell structure absorbers with complete impedance matching, feasible for optimizing the impedance matching performance of yolk-shell materials.

In this study, an N-doped honeycomb carbon shell with a CoFe_2O_4 core was synthesized by solvothermal method followed by hydrolysis, etching, and annealing. The resulting composite materials combined the advantages of magnetic materials and carbon materials to provide various absorbing mechanisms. The abundant honeycomb pores and voids between the core and shell generated an enormous amount of reflection space and specific surface area, allowing multiple reflections of electromagnetic waves and improved absorbing efficiency. The doping of N atoms reduced the electrical conductivity, introduced dipoles and defects, and improved the impedance matching on the microsphere surface. This, in turn, declined the immediate reflection of microwave incidents on the surface. The electromagnetic properties of the resulting yolk-shell structure absorbers were thoroughly discussed along with the MA mechanism of synthetic samples. Overall, these findings provide new insights into the impedance matching design of MA materials with yolk-shell structure, relevant for future uses.

2. Experimental

2.1. Materials

Cobalt(II) chloride hexahydrate ($\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$) was purchased from Xilong Scientific. Iron chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$, 99.0%) and tetrapropyl orthosilicate (TPOS, 97%) were obtained from Aladdin. Resorcinol ($\text{C}_6\text{H}_6\text{O}_2$, 99%), formaldehyde solution (CH_2O , 37.0 wt%), and polyethylene glycol 2000 (PEG 2000) were received from Macklin. Melamine ($\text{C}_3\text{H}_6\text{N}_6$, 99%), ammonia solution ($\text{NH}_3 \cdot \text{H}_2\text{O}$, 25.0 wt%), ethylene glycol (EG, 98%), urea ($\text{CH}_4\text{N}_2\text{O}$, 99%), and sodium hydroxide (NaOH, 97%) were all provided by Sinopharm. All chemicals were of AR grade and used as received without any further purification. Deionized water was prepared in our laboratory using an ultrapure water machine.

2.2. Preparation of CoFe_2O_4 nanospheres

CoFe_2O_4 nanospheres were synthesized by the hydrothermal process. Typically, $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ (928.65 mg), $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ (1688.25 mg), PEG-2000 (1000 mg), and urea (2800 mg) were dispersed in EG (80 mL) under magnetic stirring for 2 h. Next, the solution was transferred into a 150-mL Teflon reactor and heated in an oven at 190 °C for 12 h. After cooling down to room temperature, the resulting black powders were cleaned with deionized water and absolute ethanol three times, which was followed by drying under vacuum at 45 °C for 20 h to yield CoFe_2O_4 microspheres.

2.3. Preparation of $\text{CoFe}_2\text{O}_4@\text{Resorcinol-Formaldehyde (RF)}/\text{SiO}_2$ nanospheres

First, the as-prepared CoFe_2O_4 nanospheres (200 mg) were dispersed in mixed deionized water (80 mL) and absolute ethanol (15 mL) solution. The resulting mixture was then ultrasonically dispersed to separate for 10 min followed by mechanical agitation. Next, $\text{NH}_3 \cdot \text{H}_2\text{O}$ (3 mL) and TPOS (1.5 mL) were added into the mixture and the contents were kept stirring for 2 min. Then, formaldehyde solution (0.7 mL) and resorcinol (500 mg) were added dropwise into the mixture followed by stirring for 24 h at 30 °C. The obtained mixture was then dried under vacuum at 60 °C for 12 h to obtain $\text{CoFe}_2\text{O}_4@\text{RF}/\text{SiO}_2$ microspheres.

2.4. Preparation of $\text{CoFe}_2\text{O}_4@\text{HC}$ nanospheres

The collected $\text{CoFe}_2\text{O}_4@\text{RF}/\text{SiO}_2$ microspheres were fully ground followed by carbonization in a tube furnace containing Ar at 700 °C ($5^\circ\text{C}\cdot\text{min}^{-1}$) for 2 h to yield $\text{CoFe}_2\text{O}_4@\text{C}/\text{SiO}_2$. The honeycomb structure was obtained by etching the reaction. To this end, $\text{CoFe}_2\text{O}_4@\text{C}/\text{SiO}_2$ microspheres were dispersed in a solution of NaOH (1 M, 100 mL) and left under mechanical stirring for 24 h to remove SiO_2 . The resulting product was then cleaned with deionized water and absolute ethanol three times to yield $\text{CoFe}_2\text{O}_4@\text{HC}$ after drying in vacuum at 60 °C for 12 h. The honeycomb carbon shell (HC) was synthesized following the same procedure mentioned above, but without using CoFe_2O_4 as a core.

2.5. Preparation of $\text{CoFe}_2\text{O}_4@\text{HCN}$ nanospheres

$\text{CoFe}_2\text{O}_4@\text{HC}$ microspheres were ground and mixed thoroughly with melamine at a mass ratio of 2:3. Then, the mixed powders were heat treated under N_2 at 700 °C ($5^\circ\text{C}\cdot\text{min}^{-1}$) for 2 h, and the final product was obtained by magnetic separation.

2.6. Characterization

Crystalline structures and phase compositions were analyzed by X-ray diffraction (XRD, Bruker, D8 Advance) with $\text{CuK}\alpha$ radiation and $\lambda = 1.5406$ Å. The morphologies and particle sizes were characterized by scanning electron microscopy (SEM, Hitachi SU8010, including energy-dispersive spectroscopy (EDS)). The surface components and chemical bonds were examined by X-ray photoelectron spectroscopy (XPS, Escalab-250 X-ray). The Raman spectra were obtained using a Raman spectrometer (alpha-300R, Witec) at a laser excitation of 532 nm. Fourier transform infrared (FT-IR) spectrometer were measured by Thermo Scientific Nicolet iS20. The Brunauer–Emmett–Teller (BET) surface area was measured using a TriStar II 3020, Micromeritics Instrument Corporation. Barret–Joyner–Halenda (BJH) pore size distribution was calculated from the absorption isotherms by using the nonlocal density functional theory. The magnetic properties were obtained using a magnetic measurement system (VSM, Lakeshore7407).

2.7. Electromagnetic parameters measurement

Before measurements, the samples were ground into fine powders

followed by mixing with paraffin with a mass ratio (40 wt% for samples, 60 wt% for paraffin) before pressing through a specific mold into a coaxial ring with an outer diameter of 6.95 mm and inner diameter of 3.05 mm. The complex permittivity ($\epsilon_r = \epsilon' - j\epsilon''$) and permeability ($\mu_r = \mu' - j\mu''$) of each composite were evaluated using a vector network analyzer (VNA, Agilent N5244A) in the range of 2–18 GHz.

3. Results and discussion

The fabrication process of CoFe_2O_4 @HCN yolk-shell microspheres is shown in Fig. 1. First, CoFe_2O_4 microspheres were formed spontaneously from Fe^{3+} and Co^{2+} in solution by the hydrothermal reaction. Second, CoFe_2O_4 was used as the core to allow the hydrolysis of TPOS and polymerization of RF for the simultaneous formation of the inter-laced structure. Third, the RF was decomposed into carbon by annealing, and SiO_2 was subsequently etched with NaOH to obtain CoFe_2O_4 @HC. Finally, CoFe_2O_4 @HCN was obtained by annealing CoFe_2O_4 @HC and melamine mixture under N_2 .

Fig. 2a shows the SEM image of CoFe_2O_4 microspheres. Obviously, CoFe_2O_4 particles displayed a spherical shape with irregular grooves and pores (diameters around 230–310 nm). Such structure is expected to be helpful in enhancing the multi-reflection and absorption of microwaves. Fig. 2b exhibits the morphology of the as-prepared honeycomb porous carbon shells, illustrating a regular spherical shape with clear nanoscale cellular pores. Fig. 2c demonstrates that CoFe_2O_4 @HC microspheres exhibited HC with regular shell shape uniformly coated on CoFe_2O_4 core. The top right of Fig. 2c also shows a bare CoFe_2O_4 core maintaining the same morphology similar to the original CoFe_2O_4 microspheres, with the thickness of the HC shell of 65 nm being independent of the core diameter. Therefore, the coating of the honeycomb carbon shell and annealing barely affected the morphology of the CoFe_2O_4 core. The morphology of CoFe_2O_4 @HCN shown in Fig. 2d reveals that the doping of N atoms did not affect the porosity and structure of the HC shell. The HC/HCN shell consisted of numerous honeycomb pores structures, introducing interface and reflective cavity. The unique yolk-shell structure and multi-reflective interface were critical to the attenuation of microwave energy. The EDS element mapping of CoFe_2O_4 @HCN shown in Fig. 2e confirms the presence of N with the same distribution of C, indicating the successful doping of the N atoms into the material microsphere. Overall, the yolk-shell structure with numerous honeycomb pores was successfully synthesized.

Typical XRD patterns of HC, CoFe_2O_4 , CoFe_2O_4 @HC, and CoFe_2O_4 @HCN are presented in Fig. 3a. All samples associated with CoFe_2O_4 showed peaks perfectly corresponding to PDF#22–1086 (CoFe_2O_4). The diffraction peaks at 2θ values of 18.29° , 30.08° , 35.43° , 43.06° , 56.94° , and 62.53° were indexed to (111), (220), (311), (400), (511), and (440) crystal planes of CoFe_2O_4 (PDF#22–1086), respectively. The CoFe_2O_4 , CoFe_2O_4 @HC, and CoFe_2O_4 @HCN samples exhibited high crystallinity, indicating that the honeycomb carbon shell and annealing did not affect the core crystal structure. The XRD pattern of HC revealed a broad peak at 19° – 23° , indicating the presence of the amorphous carbon [30], as confirmed by the XRD patterns of CoFe_2O_4 @HC and CoFe_2O_4 @HCN, revealing the successful coating of HC and HCN on CoFe_2O_4 microspheres.

The Raman spectra of CoFe_2O_4 @HC and CoFe_2O_4 @HCN are shown

in Fig. 3b. The peak at 1350 cm^{-1} is attributed to the disordered vibrations at the boundary of the hexagonal Brillouin zone, expressing the level of defects in the lattice of carbon atoms. The peak at 1590 cm^{-1} is attributed to the stretching vibration mode of the in-plane bond of C atoms related to the graphitization degree [31,32]. The defect density of a material can be expressed by the value of the intensity ratio of peaks (I_D/I_G). The calculated I_D/I_G values of CoFe_2O_4 @HC and CoFe_2O_4 @HCN are 0.9474 and 0.9704, respectively. The elevated I_D/I_G value after nitridation indicates an increase in defects of the carbon shell caused by the doping of N atoms and the production of N-containing functional groups. These defects and functional groups produced interfacial polarization and dipole polarization effects in the presence of the alternating electromagnetic field. Moreover, the multi-scattering among the defects may contribute to the loss capacity of electromagnetic waves [13].

The magnetic performances under alternating magnetic fields of the samples are provided in Fig. 3c. Noteworthy, the magnetization of CoFe_2O_4 , CoFe_2O_4 @HC, and CoFe_2O_4 @HCN was measured by VSM under magnetic fields of $-20,000$ to $20,000$ Oe at 25°C . The presence of obvious long-narrow hysteresis loops indicates the existence of magnetic hysteresis loss. The absorbers showed strong ferromagnetic behaviors with M_s value of $72.94\text{ emu}\cdot\text{g}^{-1}$ for CoFe_2O_4 , while those of CoFe_2O_4 @HC and CoFe_2O_4 @HCN reduced to 62.40 and $65.48\text{ emu}\cdot\text{g}^{-1}$, respectively. The decrease in M_s value can be explained by the increase in the content of the non-magnetic components. Furthermore, the H_c values of CoFe_2O_4 , CoFe_2O_4 @HC, and CoFe_2O_4 @HCN were estimated to be 93, 142, and 152 Oe, respectively (Fig. 3c). The increase in H_c values was attributed to the enhanced anisotropy caused by annealing [33]. In general, high initial magnetic permeability (μ_i) favors strong magnetic losses, and μ_i can be calculated according to Eq. (1) [34]:

$$\mu_i = \frac{M_s^2}{akH_cM_s + b\lambda\xi} \quad (1)$$

where a and b are two constants determined by the composition of materials, λ refers to the magnetostriction constant, and ξ denotes the elastic strain parameter of the material. Notably, higher M_s and lower H_c induce higher μ_i , conducive to promoting stronger magnetic loss.

The surface composition and chemical bond of CoFe_2O_4 @HCN characterized by XPS are displayed in Fig. 3(d–f). The survey spectrum of CoFe_2O_4 @HCN displayed a composite mainly containing C, N, O, Fe, and Co elements (Fig. 3d). The absence of Co and Fe signals was probably caused by the effective detection depth of XPS, which did not surpass 10 nm, thereby becoming incapable of penetrating the honeycomb carbon shell with thickness of 60 nm [35]. The high-resolution spectrum of C 1s shown in Fig. 3e can be deconvoluted into four peaks ascribed to the C–C/C=C group (284.80 eV), C–N group (285.82 eV), C–O group (286.72 eV), and O–C=O group (289.06 eV) [4,8]. Fig. 3f exhibits that the high-resolution N 1s spectrum can be deconvoluted into four peaks, attributed to pyridinic N (398.57 eV), pyrrolic N (399.87 eV), graphitic N (400.97 eV), and oxidized N (403.29 eV), respectively [36,37]. This supported the EDS data and confirmed the successful incorporation of N atoms into the honeycomb carbon shell. The N atoms with different electronegativities than C atoms generated the dipole moments. The deflection of dipole moment under alternating electromagnetic field benefited the loss in electromagnetic waves.

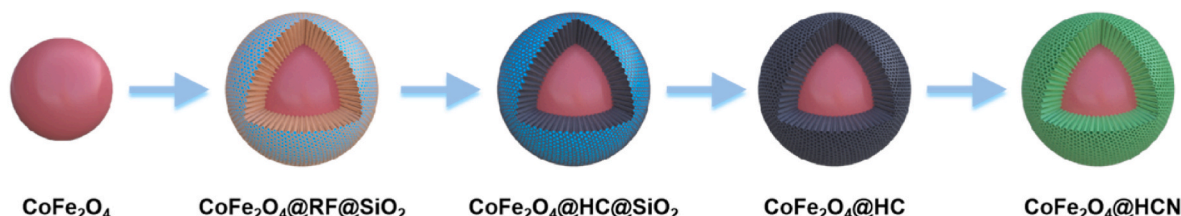


Fig. 1. The preparation process of CoFe_2O_4 @HCN microspheres.

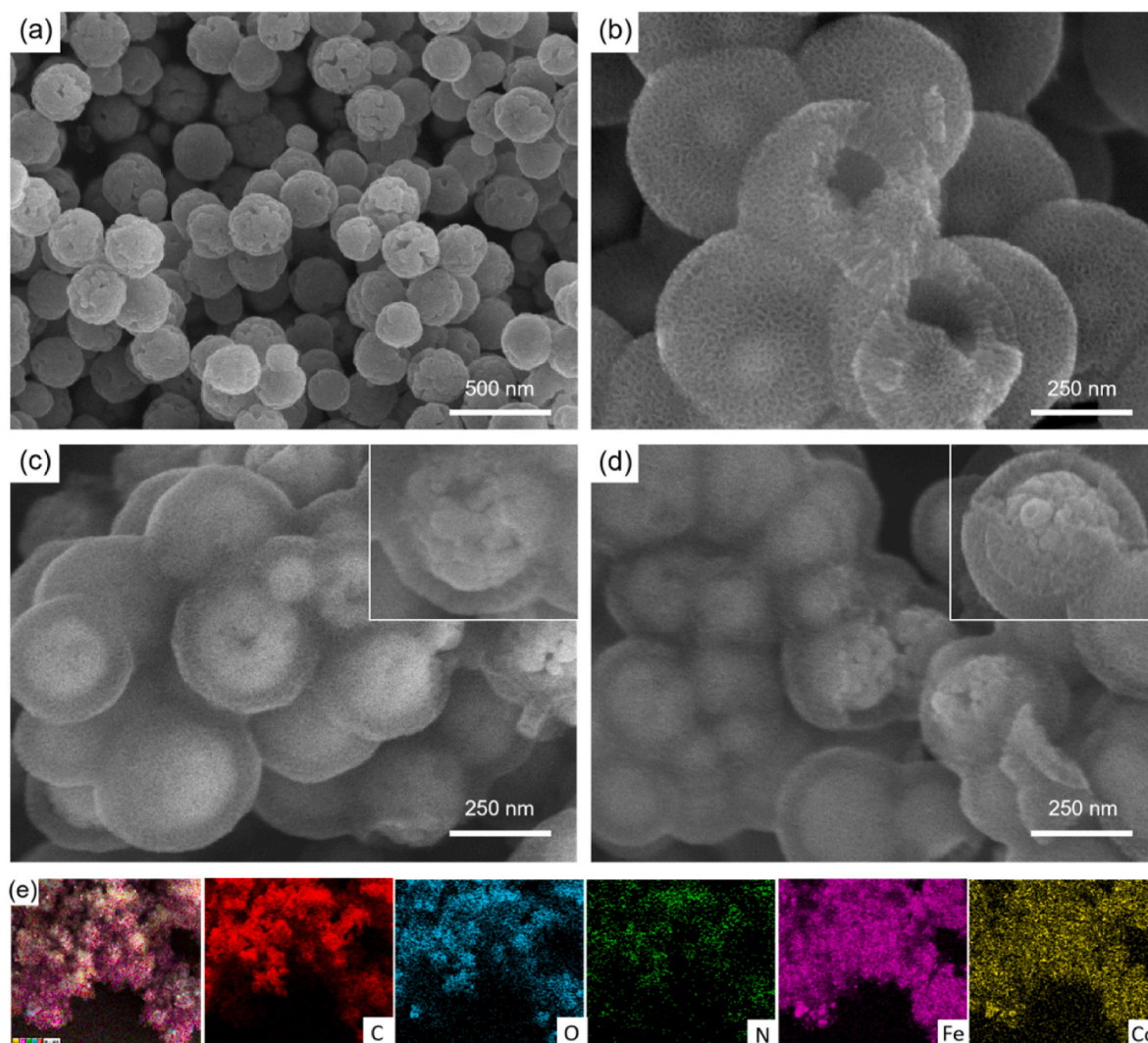


Fig. 2. SEM images of (a) CoFe_2O_4 , (b) HC, (c) $\text{CoFe}_2\text{O}_4@\text{HC}$, (d) and $\text{CoFe}_2\text{O}_4@\text{HCN}$. (e) EDS mapping of $\text{CoFe}_2\text{O}_4@\text{HCN}$.

Fig. S1 displays the Fourier transform infrared (FTIR) spectrum of $\text{CoFe}_2\text{O}_4@\text{HCN}$. The absorption peaks at 1629, 1388, and 1063 cm^{-1} are attributed to the C=C stretching vibration, =C–H in-plane vibration, and deformation vibrations of C–H, respectively [20,22]. The peak at 1459 cm^{-1} represents conjugated C–N stretching [22,38]. The peak appearing at 601 cm^{-1} is associated with Fe–O [39], indicating the existence of CoFe_2O_4 . The FTIR spectroscopy result shows the consistency with the XRD and XPS results.

The microporous morphologies of the materials were analyzed by N_2 adsorption isotherms. The N_2 sorption isotherms and BJH pore size distributions of HC, $\text{CoFe}_2\text{O}_4@\text{HC}$, and $\text{CoFe}_2\text{O}_4@\text{HCN}$ are presented in Fig. 4. All three materials exhibited type IV isotherms, indicating characteristics mesoporous morphologies [26,40]. The adsorption capacities of HC increased sharply in the low-pressure region ($P/P_0 < 0.1$), indicating the presence of a microporous structure corresponding to the smaller diameter pores located in the inner wall. HC showed an H4-type hysteresis loop at P/P_0 of 0.4–1.0 caused by the capillary condensation phenomenon, revealing the presence of layered honeycomb holes in the materials and corroborating the SEM data. Fig. 4(b and c) illustrate that both $\text{CoFe}_2\text{O}_4@\text{HC}$ and $\text{CoFe}_2\text{O}_4@\text{HCN}$ present a slow increase in adsorption capacity ($P/P_0 < 0.1$), which can be attributed to the blocking of the micropores by internal CoFe_2O_4 and partial destruction caused by annealing. $\text{CoFe}_2\text{O}_4@\text{HC}$ and $\text{CoFe}_2\text{O}_4@\text{HCN}$ revealed an H3-type hysteresis loop at P/P_0 of 0.4–1.0. The rapid increase in

adsorption at P/P_0 of 0.8–1.0 was contributed by the accumulation gap of particles. The surface grooves of CoFe_2O_4 nanospheres and voids between the nanospheres and carbon shells were responsible for the adsorption effect. In BJH model, the pore size of small holes can be characterized by $dV/d\log(D)$. The BJH pore size distributions of HC shown in Fig. 4a reveal the presence of peaks at 2.3 nm, corresponding to the honeycomb mesopores of the carbon shell. The large pores at 52 nm in $\text{CoFe}_2\text{O}_4@\text{HC}$ corresponded to the grooves in the CoFe_2O_4 cores and the gaps between the broken shells and cores (Fig. 4b). The macropore size of $\text{CoFe}_2\text{O}_4@\text{HCN}$ further increased to 72 nm (Fig. 4c), consistent with the N_2 adsorption/desorption curves. For porous materials, the presence of mesopores and macropores provide numerous surfaces for the scatter and reflection of microwaves, contributing extremely to the MA properties of absorbers. The BET surface areas of HCN, $\text{CoFe}_2\text{O}_4@\text{HC}$, and $\text{CoFe}_2\text{O}_4@\text{HCN}$ were 886.77, 677.17, and $592.27\text{ m}^2\cdot\text{g}^{-1}$, respectively. The decrease in the values of $\text{CoFe}_2\text{O}_4@\text{HC}$ and $\text{CoFe}_2\text{O}_4@\text{HCN}$ stemmed from the addition of CoFe_2O_4 core, thereby masking the internal pores. The annealing and oxidation also caused the collapse of some microporous structures. High BET surface areas and nanoscale pore size indicate the material with elevated porosity. The Maxwell Garnet equation model and air-to-material interface intrinsic reflection coefficient (I) can be calculated according to Eqs. (2) and (3) [41–43].:

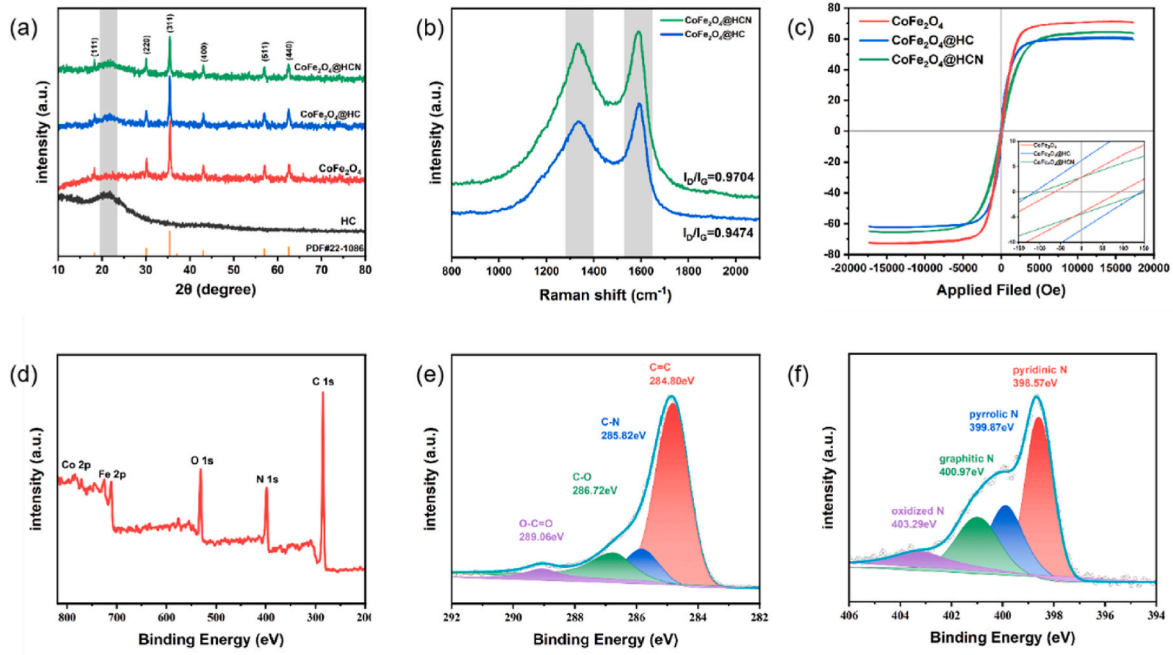


Fig. 3. (a) XRD patterns of HC, CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN. (b) Raman spectra of CoFe₂O₄@HC and CoFe₂O₄@HCN. (c) Magnetic hysteresis loops of CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN. XPS spectra of (d) survey spectrum, (e) C 1s, and (f) N 1s.

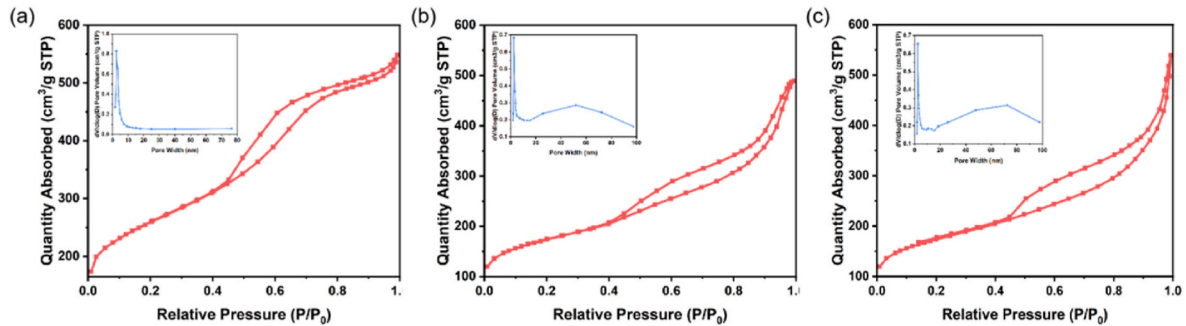


Fig. 4. N₂ adsorption–desorption isotherm and pore size distribution of (a) HC, (b) CoFe₂O₄@HC, and (c) CoFe₂O₄@HCN.

$$\epsilon_{eff}^{MG} = \left[\frac{(2\epsilon_1 + \epsilon_2) + 2\nu(2\epsilon_2 - \epsilon_1)}{(2\epsilon_1 + \epsilon_2) - \nu(2\epsilon_2 - \epsilon_1)} \right] \epsilon_1 \quad (2)$$

$$\Gamma = \frac{\sqrt{\frac{\mu_r}{\epsilon_r}} - 1}{\sqrt{\frac{\mu_r}{\epsilon_r}} + 1} \quad (3)$$

where ν is the volume fraction of the pores. ϵ_1 and ϵ_2 represent the permittivity of matrix material and air in the pores. ϵ_r and μ_r represent the complex permittivity and complex permeability, respectively.

Large air volume facilitates the reduction of effective permittivity. For materials with low magnetic permeability, smaller permittivity values result in lower Γ , leading to the diminished reflection of the microwaves incident on the materials. Moreover, the honeycomb 3D carbon structure could also form conductive networks. According to the literature [44], electrons in electrons hopping mode could transfer in the yolk-shell structure consisting of CoFe₂O₄ and carbon by absorbing microwave energy to transform it into heat through dissipation.

To evaluate the MA properties, the reflection loss (RL) was calculated by using the transmission line theory [45,46]:

$$Z_{in} = Z_0 \sqrt{\mu_r/\epsilon_r} \tan h [j(2\pi f d / c) \sqrt{\mu_r \epsilon_r}] \quad (4)$$

$$RL = 20 \log |(Z_{in} - Z_0) / (Z_{in} + Z_0)| \quad (5)$$

where Z_{in} is the input characteristic impedance of the materials, Z_0 represents the impedance of free space, c refers to the velocity of light in vacuum, and d is the thickness of material.

The 3D diagrams and corresponding contour diagrams of the RL values of HC, CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN at frequencies within 2–18 GHz and d values of 1–5.5 mm are provided in Fig. 5. In general, RL below –10 dB indicates that absorbing rates of electromagnetic waves reach 90%, while RL below 20 dB refers to absorbing rates reaching 99%. Furthermore, absorbers with an RL of –10 dB or less possess practical application capability [47]. Fig. 5a illustrates that the hollow HC exhibits low reflection loss characteristics with RL_{min} value of –20.57 dB at 10.96 GHz for a thickness of 2.5 mm and the EAB is 3.56 GHz (9.28–12.84 GHz). For bare CoFe₂O₄ microspheres (Fig. 5b), the RL_{min} at the thickness of 5.0 mm reached –17.51 dB at 6.12 GHz and EAB reached 2.52 GHz (4.92–7.44 GHz). Therefore, RL_{min}, EAB, and thickness of both materials were not sufficient for practical application. Such poor MA properties were consistent with their monotonic composition and structure. In contrast, the materials after the composite formation exhibited excellent MA properties. Fig. 5c

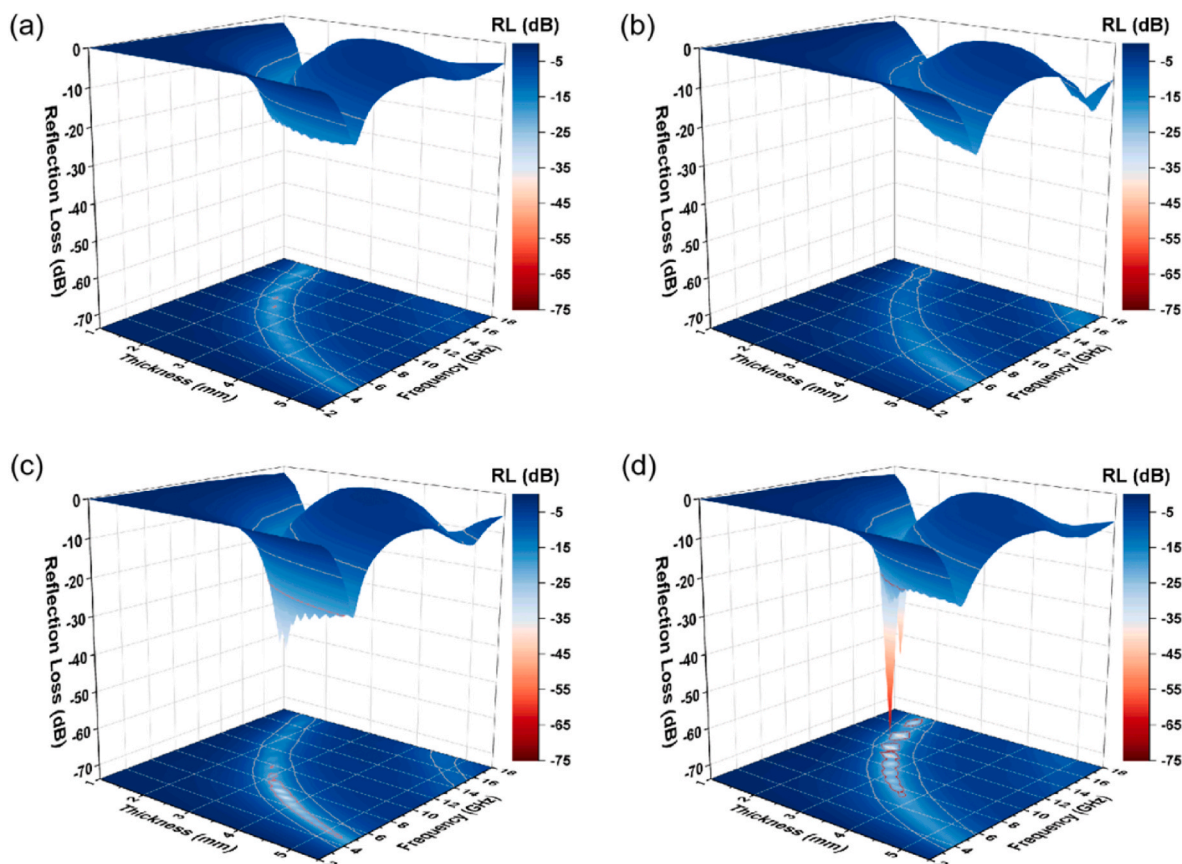


Fig. 5. Three-dimensional reflection loss of (a) HC, (b) CoFe_2O_4 , (c) CoFe_2O_4 @HC, and (d) CoFe_2O_4 @HCN.

demonstrates that the RL_{\min} value of CoFe_2O_4 @HC at d of 3.5 mm reached -34.49 dB at 7.32 GHz, while EAB reached 2.96 GHz (from 5.96 to 8.92 GHz). The absorber exhibits useable absorbing properties at 2–5 mm. In comparison, the other RL_{\min} values were much lower than -10 dB. Thus, the unique structure formed by CoFe_2O_4 and HC and synergistic effect of magnetic and conductive properties achieved decent MA properties. Among all four samples, CoFe_2O_4 @HCN (Fig. 5d) illustrated the most desirable characteristics suitable for applications as an excellent absorber. The RL_{\min} value at 2.5 mm thickness reached -68.03 dB at 11.36 GHz, and the EAB was 4.52 GHz (from 9.4 to 13.92 GHz). Significantly, with the decrease in the thickness to 1.99 mm, the RL_{\min} value still maintained -58.34 dB, while the EAB broadened to 5.92 GHz (from 11.92 to 17.84 GHz). Therefore, the doping of N atoms effectively improved the MA properties and thinned the matching thickness of the samples. The MA performance of typical yolk-shell absorbers with binary or more materials and absorbers with CoFe_2O_4 and carbon based are listed in Table 1. The results indicate that CoFe_2O_4 @HCN possesses a high reflection loss ability and wide EAB with lower thickness, in addition, CoFe_2O_4 @HCN owns a lighter density compared to extreme performance absorbers, resulting a superior applicability.

In general, the main factor determining the MA performance consists mainly of impedance matching ($|Z_{\text{in}}/Z_0|$) [48], where a value of $|Z_{\text{in}}/Z_0|$ closer to 1 represents a better impedance matching and lower proportion of reflection. The changes in curves of $|Z_{\text{in}}/Z_0|$ values versus frequency at different thicknesses are presented in Fig. 6. The $|Z_{\text{in}}/Z_0|$ values of HC and CoFe_2O_4 exhibit significant deflection from the dashed line ($|Z_{\text{in}}/Z_0| = 1$), as shown in Fig. 6a and b. For CoFe_2O_4 @HC (Fig. 6c), the $|Z_{\text{in}}/Z_0|$ value is close to the dashed line at 6.28 GHz with a thickness of 3.5 mm, which can be attributed to the combination of CoFe_2O_4 and HC. Fig. 6d exhibits that CoFe_2O_4 @HCN show two regions close to the dashed line at 11.36 GHz at 2.5 mm and 14.4 GHz at 1.99 mm. Thus, the introduction of N atoms led to significant optimization of the impedance

Table 1

Microwave absorption performance comparison of yolk-shell absorbers/ CoFe_2O_4 & carbon based absorbers in recent literature.

Absorbers	RL_{\min} (dB)	EAB (GHz)	Thickness (mm)	Refs.
MnO_2 @PANI	-52.06	4.56	3.00	[55]
Fe_3O_4 @C@ MnO_2	-58.25	5.56	4.20	[65]
$\text{Co}_3\text{Fe}_7/\text{FeO@C}$	-57.60	5.27	2.50	[66]
CIP@void@NC	-25.70	6.90	1.70	[67]
ZnFe_2O_4 @PHCMS	-51.43	3.52	4.80	[68]
NRGO/hollow CoFe_2O_4	-44.70	5.20	1.80	[69]
rGO/SiC/ CoFe_2O_4	-65.49	2.30	1.06	[70]
CoFe_2O_4 /MWCNTs	-50.80	<3.36	4.20	[71]
CoFe_2O_4 , (Fe, Ni) and carbon frame	-34.30	<6.55	2.00	[72]
CNTs@ CoFe_2O_4 /polyimide	-54.40	6.70	2.60	[73]
porous carbon@ CoFe_2O_4	-67.04	5.52	2.00	[74]
CoFe_2O_4 @HCN	-68.03	4.52	2.50	This work
	-58.34	5.92	1.99	This work

matching performance of the outer layer of the absorber, indicating that the microwave's reflection would be rare at the first moment of incidence at the surface.

Since thickness exhibits an obvious influence on the MA process, the relationship between RL and thickness was investigated as a function of frequency. Fig. 6 depicts that the RL_{\min} peaks shifted toward the low-frequency region with the increase in the thickness of the materials, which is explained by the 1/4 wavelength matching model. The matching thickness (t_m) and peak frequency (f_m) satisfy the following relationship [49–51]:

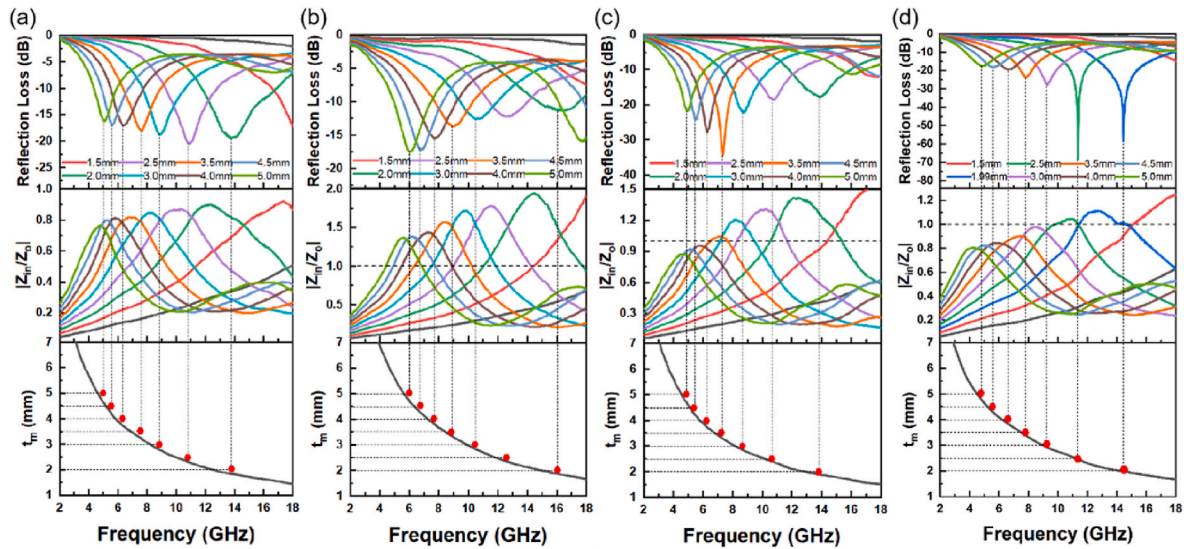


Fig. 6. Stack plots of two-dimensional reflection loss, impedance matching ($|Z_{in}/Z_0|$), and matching thickness (t_m) versus frequency for (a) HC, (b) CoFe_2O_4 , (c) $\text{CoFe}_2\text{O}_4@HC$, and (d) $\text{CoFe}_2\text{O}_4@HCN$.

$$t_m = \frac{n\lambda}{4} = \frac{nc}{4f_m \sqrt{|\epsilon_r||\mu_r|}} \quad (n = 1, 3, 5 \dots) \quad (6)$$

Notably, for an absorber conforming to the 1/4 wavelength matching model, the reflected electromagnetic waves from the interface lag the incident wave by 180° , resulting in a strong interference effect leading to high RL_{min} values [29,52]. Fig. 6 exhibits that the experimental matching thicknesses of all samples highly matched the theoretically calculated values, confirming absorbers obeying the 1/4 wavelength model.

In order to further analyze the MA mechanism of the as-prepared samples, the ϵ' , ϵ'' , μ' , and μ'' of HC, CoFe_2O_4 , $\text{CoFe}_2\text{O}_4@HC$, and $\text{CoFe}_2\text{O}_4@HCN$ were obtained between 2 and 18 GHz (Fig. 7). Herein, the ϵ' and ϵ'' represent the storage capacity and loss capacity of electric energy, respectively, while μ' and μ'' refer to the storage capacity and

loss capacity of magnetic energy, respectively. The variations in ϵ' of all samples versus frequency are provided in Fig. 7a. Obviously, the ϵ' values of three samples containing HC are higher than those of pure CoFe_2O_4 due to the electrical conductivity of HC. The ϵ' values of $\text{CoFe}_2\text{O}_4@HC$ and $\text{CoFe}_2\text{O}_4@HCN$ reduced when weakly conducting components were added. For $\text{CoFe}_2\text{O}_4@HCN$, the doping of N atoms restrained the conductivity of the carbon shell, resulting in a lower ϵ' when compared to $\text{CoFe}_2\text{O}_4@HC$ [18], and lower ϵ' was found to be conducive to improving the dielectric loss tangent ($\tan\delta_e$). The ϵ'' values versus frequency are presented in Fig. 7b. The ϵ'' can be drawn according to Eq. (7) [18]:

$$\epsilon'' = \epsilon_r'' + \epsilon_c'' = \frac{\epsilon_s - \epsilon_\infty}{1 + \omega^2\tau^2} + \frac{\sigma}{\omega\epsilon_0} \quad (7)$$

where ϵ_r'' and ϵ_c'' , successively, represent the polarization loss and

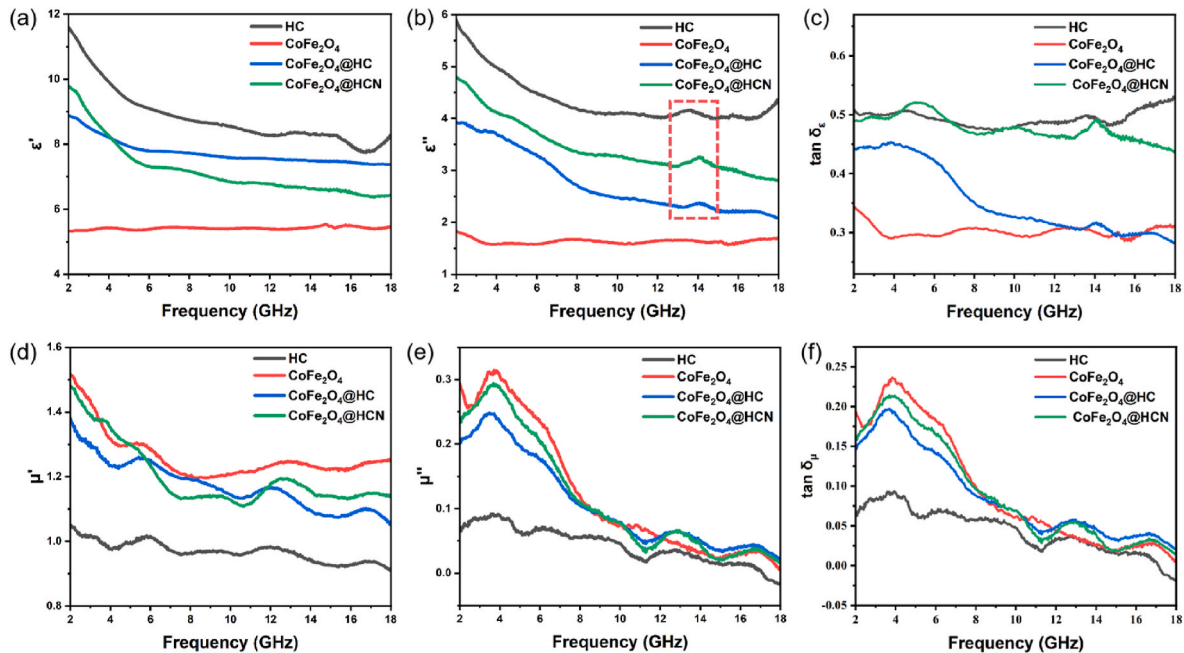


Fig. 7. (a) Real part of permittivity, (b) imaginary part of permittivity, (c) dielectric loss tangents $\tan \delta_e$, (d) real part of permeability, (e) imaginary part of permeability, and (f) magnetic loss tangents $\tan \delta_m$, of HC, CoFe_2O_4 , $\text{CoFe}_2\text{O}_4@HC$, and $\text{CoFe}_2\text{O}_4@HCN$.

conductive loss, while ϵ_s , ϵ_∞ , σ , and τ correspond to the static dielectric constant, dielectric constant at infinite frequency, electrical conductivity, and polarization relaxation time, respectively [53,54].

The ϵ'' of HC seemed to be relatively large due to the highest electrical conductivity. CoFe₂O₄@HC led to the decrease in ϵ'' due to the addition of insulating CoFe₂O₄. Noteworthy, after doping with N atoms, the ϵ'' of CoFe₂O₄@HCN significantly increased when compared to CoFe₂O₄@HC owing to the stronger polarization loss performance, in particular, for dipole polarization generated by C–N dipoles. All three materials containing similar carbon shells showed a frequency relaxation resonance peak at 13–14 GHz associated with the interface polarization effect of honeycomb carbon shells [55,56]. Usually, $\tan\delta_e$ ($\tan\delta_e = \epsilon''/\epsilon'$) is used to evaluate the dielectric loss capacity of the absorber. The $\tan\delta_e$ of HC and CoFe₂O₄@HCN was much larger than those of CoFe₂O₄@HC and CoFe₂O₄. The high ϵ_r of HC was not conducive to the impedance matching, while CoFe₂O₄@HCN achieved optimal dielectric properties due to the lower ϵ' and synergistic effect issued from conductive loss, interfacial polarization loss, and dipole polarization loss.

The mechanism of dielectric loss expressed by Debye polarization relaxation is divided into electron polarization, interface polarization, ion polarization, and dipole polarization. At high frequencies, current research exhibits the occurrence of both ion polarization and electron polarization. The interface polarization and dipole polarization are the main polarization mechanisms at 2–18 GHz. The Debye relaxation theory can be described by using Eq. (8) [57–59]:

$$\left(\epsilon' - \frac{\epsilon_s + \epsilon_\infty}{2}\right)^2 + (\epsilon'')^2 = \left(\frac{\epsilon_s - \epsilon_\infty}{2}\right)^2 \quad (8)$$

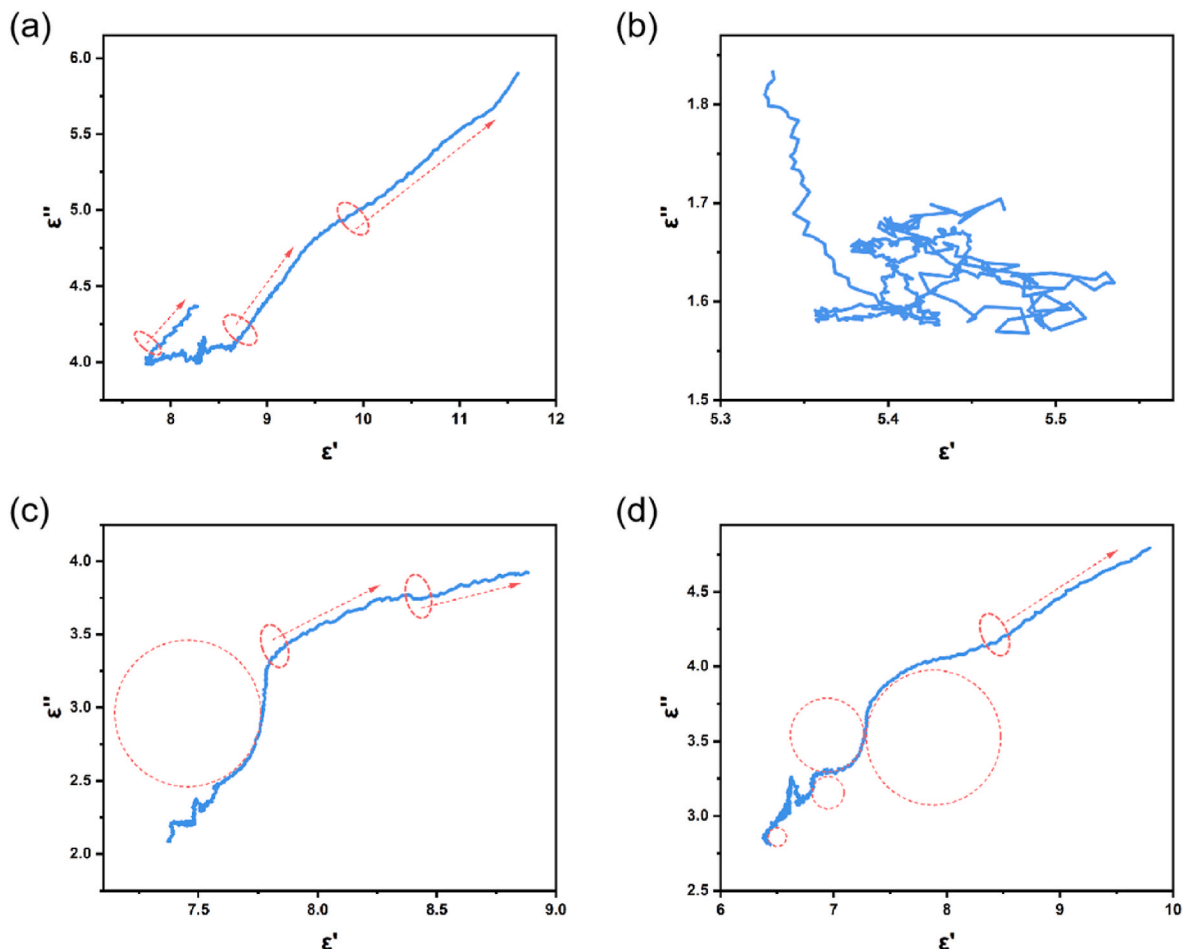


Fig. 8. Cole–Cole semicircles (ϵ' – ϵ'' curves) of (a) HC, (b) CoFe₂O₄, (c) CoFe₂O₄@HC, and (d) CoFe₂O₄@HCN.

The semicircle relationship between ϵ' and ϵ'' (Cole–Cole semicircle) represents a Debye relaxation process. According to the previous study [11], the conductive loss replaces the dielectric loss as the main loss mechanism for ϵ' – ϵ'' curve showing a straight line [52,60]. The ϵ' – ϵ'' curves of HC, CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN are presented in Fig. 8. HC presented three-part linear relationship curves in Fig. 8a, indicating that loss capacity was mainly caused by conductive loss issued from the excellent electrical conductivity and 3D conductive network. Fig. 8b exhibits the presence of a completely disordered curve, indicating no conductive loss or dielectric loss in CoFe₂O₄. Fig. 8c presents that CoFe₂O₄@HC displayed one Cole–Cole semicircle and two-part linear relationship curves. The Cole–Cole semicircle corresponded to the Debye relaxation process caused by the heterogeneous interface of CoFe₂O₄ and HC. The heterogeneous aggregation of local charges around the interface formed interfacial polarization under the alternating electromagnetic field [26]. For CoFe₂O₄@HCN, four semicircles and one linear part were observed (Fig. 8d), indicating the occurrence of conductive loss and multi-relaxation processes such as interfacial polarization and dipole polarization. The natural dipole moment formed by C atoms and N atoms rearranged and oriented along the direction of the electric field under the action of the alternating electric field by absorbing electromagnetic energy [29].

The changes in μ' values as a function of frequency are presented in Fig. 7d. Noteworthy, the change is theoretically related to the proportion of magnetic components. The μ' values versus frequency of CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN fluctuated between 1.1 and 1.4, while the μ' of HC was below 1.0. The samples CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN showed high μ'' values than HC, as presented in Fig. 7e. The μ'' of CoFe₂O₄ was the highest in the range of

2–8 GHz, while that of CoFe₂O₄@HCN was slightly lower than that of CoFe₂O₄ but slightly higher than that of CoFe₂O₄@HC. This may be related to the annealing process, which induced changes in M_s and H_c of CoFe₂O₄ core. Consequently, the μ'' curves of CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN almost overlapped in the region of 8–18 GHz. The magnetic loss capacity was evaluated by $\tan\delta_\mu$ ($\tan\delta_\mu = \mu''/\mu'$), and the frequency-dependent $\tan\delta_\mu$ of as-prepared samples are provided in Fig. 7f. The trend of the $\tan\delta_\mu$ curves was almost the same as that of μ'' curves as a function of frequency. Samples containing CoFe₂O₄ displayed a higher magnetic loss than pure HC.

The mechanism of magnetic loss for the absorber can mainly be segmented into eddy current loss, domain-wall resonance, natural resonance, and magnetic hysteresis loss [53]. The domain-wall resonance of materials appeared only at the frequency of 1–100 MHz [61]. The natural resonance effect can be calculated by using Eqs. 9–11 [49, 62]:

$$K = \mu_0 M_s H_c / 2 \quad (9)$$

$$H_a = 4|K|/3\mu_0 M_s \quad (10)$$

$$2\pi f_r = rH_a \quad (11)$$

where K represents anisotropy coefficient, μ_0 represents the permeability of vacuum, H_a is the anisotropy energy, f_r denotes the resonant frequency, and r is the gyromagnetic ratio.

Herein, the resonant frequency depends on the anisotropy of the material mainly manifested by H_c . Fig. 3c exhibits that the H_c values of CoFe₂O₄@HC and CoFe₂O₄@HCN were larger than that of pure CoFe₂O₄, owing to the high natural resonance properties. The curve enclosed with a certain area by the hysteresis lines during magnetization indicates the existence of hysteresis loss. The eddy current loss coefficient could be described by using Eq. (12) [59,63]:

$$C_0 = \mu'' (\mu')^{-2} f^{-1} = 2\pi\mu_0 d^2 \delta \quad (12)$$

where C_0 represents the eddy current loss coefficient, d is the diameter of the nanoparticle, and δ refers to the electrical conductivity.

The occurrence of eddy current loss in the absorber should result in a constant C_0 value versus frequency. The C_0 curves for HC, CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN shown in Fig. 9a reveal fluctuating C_0 values for all samples as a function of frequency, indicating minimal eddy current loss. Thus, the magnetic loss was mainly derived from natural resonance and magnetic hysteresis loss.

Attenuation constant (α) is used to evaluate the attenuation capacity of materials, and as a crucial parameter useful to determine the MA performance, which can be described in terms of Eq. (13) [64], as

follows:

$$\alpha = \frac{\sqrt{2}\pi f}{c} \times \sqrt{(\mu''e'' - \mu' \epsilon') + \sqrt{(\mu''e'' - \mu' \epsilon')^2 + (\mu' \epsilon'' + \mu'' \epsilon')^2}} \quad (13)$$

For impedance matching, higher α values should lead to stronger absorbing ability. The α values of four absorbers are provided in Fig. 9b. HC and CoFe₂O₄@HCN generated the highest loss abilities. For CoFe₂O₄@HCN, the high α values were issued from the synergistic effect contributed by conduction loss, dielectric loss, and magnetic loss.

Based on the above-mentioned analyses, a logical MA mechanism was constructed for CoFe₂O₄@HCN microspheres. In this mechanism, a strong ferromagnetic CoFe₂O₄ core provided ideal magnetic losses such as natural resonance and hysteresis losses. Numerous interfaces arising from the honeycomb porous structure then generated interfacial polarization loss. The doping of N atoms led to the increase in the number of dipole and defect sites regarded as polarization centers under electromagnetic fields, suitable for introducing large amounts of defective polarization and dipole polarization. Furthermore, the 3D honeycomb conductive structure of the shell layer contributed to conductive losses. The superior MA performance was issued from numerous loss mechanisms and high attenuation capability combined with excellent impedance matching benefited from the synergistic effect of the magnetic core and conductive shell, in particular, the improvement of impedance matching on the shell.

4. Conclusions

Porous honeycomb yolk-shell absorber with enhanced impedance matching, core diameter of 230–310 nm, shell thickness of 65 nm, honeycomb pore size of 2.3 nm, and doped nitrogen atoms in the carbon shell was successfully prepared. The MA properties indicated an attractive RL_{min} value (−68.03 dB) of CoFe₂O₄@HCN microspheres at 2.5 mm. The absorber showed a broad EAB through the X and Ku bands of 4.52 GHz (change from 9.4 to 13.92 GHz). The RL_{min} value maintained −58.34 dB with EAB broadening to 5.92 GHz (from 11.92 to 17.84 GHz) at 1.99 mm. The improved impedance matching and attenuation mechanisms of N-doped carbon revealed decreased effective dielectric constant, interfacial polarization, dipole polarization, and defective polarization, among others. In sum, yolk-shell structures with improved impedance matching are provided as a feasible strategy for the synthesis of future N-doped carbon materials.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence

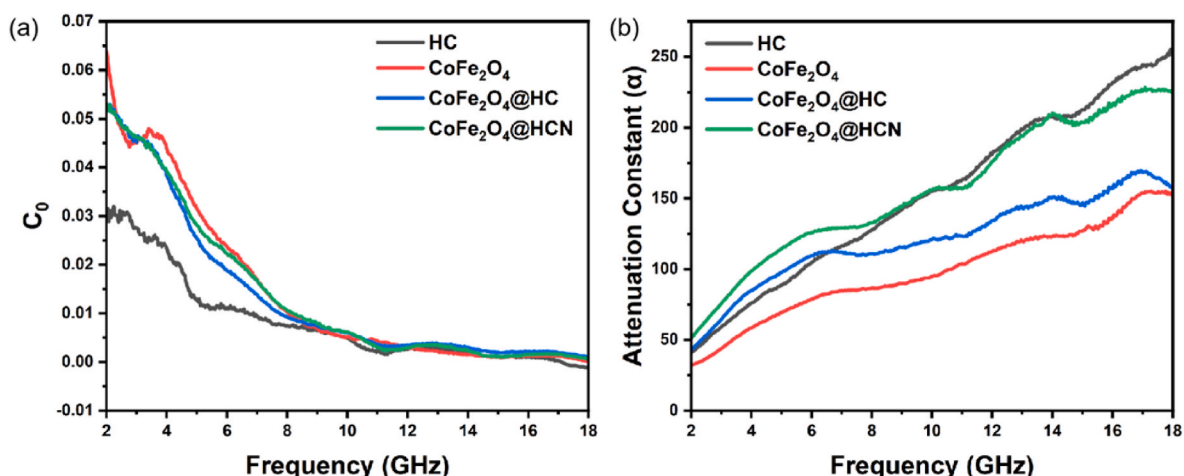


Fig. 9. (a) Eddy current loss coefficient and (b) attenuation constant (α) of HC, CoFe₂O₄, CoFe₂O₄@HC, and CoFe₂O₄@HCN.

the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.ceramint.2023.03.297>.

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