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Near-Infrared-to-Near-Infrared Optical Thermometer BaY₂O₄: Yb³⁺/Nd³⁺ Assembled with Photothermal Conversion Performance

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ABSTRACT: Nowadays, the construction of photothermal therapy (PTT) agents integrated with real-time thermometry for cancer treatment in deep tissues has become a research hotspot. Herein, an excellent photothermal conversion material, BaY_2O_4 : Yb^{3+}/Nd^{3+} , assembled with real-time optical thermometry is developed successfully. Ultrasensitive temperature sensing is implemented through the fluorescence intensity ratio of thermally coupled Nd^{3+} : 4F_j (j = 7/2, 5/2, and 3/2) with a maximal absolute and relative sensitivity of 68.88 and 3.29% K⁻¹, respectively, which surpass the overwhelming majority of the same type of thermometers. Especially, a thermally enhanced Nd^{3+} luminescence with a factor of 180 is detected with irradiation at 980 nm, resulting from the improvement in phonon-assisted energy transfer efficiency. Meanwhile, the photothermal conversion



performance of the sample is excellent enough to destroy the pathological tissues, of which the temperature can be raised to 319.3 K after 180 s of near-infrared (NIR) irradiation with an invariable power density of 13.74 mW/mm². Besides, the NIR emission of Nd³⁺ can reach a depth of 7 mm in the biological tissues, as determined by an ex vivo experiment. All the results show the potential application of BaY₂O₄: Yb³⁺/Nd³⁺ as a deep-tissue PTT agent simultaneously equipped with photothermal conversion and temperature sensing function.

INTRODUCTION

In the 21st century, cancer has become one of the most serious public health problems faced by mankind. Clinical treatment methods such as surgical operations, chemotherapy treatments, and radiation therapy are inefficient due to their scarcity of selectivity to tumor cells or high toxicity and side effects. To overcome the abovementioned problems, photothermal therapy (PTT) is becoming a hot topic in scientific research, which refers to eliminating tumor cells through the thermal effect resulting from the energy precipitation in pathological tissues in some way. During the actual treatment process, the temperature of the lesion area should be controlled in a reasonable range to guarantee not only the effective ablation of the cancer cells but also the avoidance of damaging the surrounding normal tissues caused by the overheating effect.¹⁻⁵ Therefore, it is quite essential to develop an efficient and accurate method for temperature detection to serve PTT.

The temperature sensing behavior realized by the fluorescence intensity ratio (FIR) of thermal-coupled levels (TCLs) in rare earth ion-doped upconversion (UC) materials has the advantages of noncontact, high accuracy, short response time, large spatial resolution, and strong electromagnetic interference resistance, making it eminently suitable as a temperature sensor for the PTT process.^{6–11} However, the current studies involved with FIR-based temperature measurement mainly focus on the green emission originating from the Er^{3+} : ${}^{2}H_{11/2}/{}^{4}S_{3/2}$ level, which owns excellent thermal coupling properties and temperature sensing performance.^{12–16} Never-

theless, the shallow penetration depth of green light in the biological tissues caused by strong scattering and absorption phenomena seriously restricts the clinical application of Er³⁺based thermometric probes.^{17–21} By comparison, the thermally coupled Nd³⁺: ${}^{4}F_{i}$ (*j* = 7/2, 5/2, and 3/2) levels may be a perfect replacer for FIR-based temperature sensing, which can generate far-red and near-infrared (NIR) luminescence just by locating in the optical window of the biological tissue under the sensitization of Yb^{3+} .^{14,22–26} Meanwhile, the much larger energy gap (ΔE) of the Nd³⁺: ${}^{4}F_{7/2}/{}^{4}F_{3/2}$ level (~2000 cm⁻¹) than that of the Er^{3+} : ${}^{2}H_{11/2}/{}^{4}S_{3/2}$ level (~800 cm⁻¹) means that a much higher thermometric sensitivity can be achieved. In addition, the abundant metastable energy levels with small ΔE existing in Nd³⁺ ions can effectively convert part of the absorbed NIR light into thermal energy through nonradiative relaxation (NR) processes.²⁷ Therefore, it can be anticipated that the Yb³⁺/Nd³⁺-codoped system is an ideal PTT agent, integrating outstanding photothermal conversion ability with sensitive temperature detection operating in the biological window.²⁸⁻³²

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In the present paper, BaY₂O₄ is employed as the matrix material for doping Yb³⁺ and Nd³⁺, which is an excellent host for the luminescence of rare earth ions and has good thermal stability.^{33,34} An ultrasensitive temperature sensing along with the function of photothermal conversion is found in BaY₂O₄: Yb³⁺/Nd³⁺. The thermally enhanced mechanisms of Nd³⁺ luminescence along with the temperature sensing properties realized by the thermally coupled Nd³⁺: ⁴F_j (j = 7/2, 5/2, and 3/2) levels are investigated meticulously. Meanwhile, the photothermal conversion capacity of the synthesized material with the irradiation of NIR light is studied in detail. Beyond that, a simple ex vivo examination is performed to explore the penetration depth of the sample in the biological organism. All the data indicated that BaY₂O₄: Yb³⁺/Nd³⁺ can be used as a new-type PTT agent possessing the ability of real-time temperature sensing.

EXPERIMENTAL SECTION

Chemicals. $BaCO_3$ (99.99%), Y_2O_3 (99.99%), Yb_2O_3 (99.99%), and Nd_2O_3 (99.99%) were supplied by Sinopharm Chemical Reagent Company of China, and all the chemical reagents were employed as the raw materials with no further purification.

Preparation. First, weigh the raw materials in stoichiometric proportion and then grind them in an agate mortar for 30 min. Second, the powder is transferred to a crucible with a lid for presintering in the muffle furnace at 600 degree centigrade for 6 h. Until cooled to the ambient temperature, the raw materials are taken out for a 30 min grind again. Finally, the powder is calcined for 6 h in the muffle furnace at 1300 degree centigrade.

Characterization. An XD-2 diffractometer produced by Beijing Persee is used to collect the powder X-ray diffraction (XRD) data. The spectroscopic data is detected using a FLS1000 model spectrometer made by Edinburgh Instruments Ltd. Meanwhile, a semiconductor laser with a wavelength of 980 nm is utilized as the excitation source for spectroscopic measurements. An HFS600E-PB2 temperature control device supplied by Linkam Scientific Instruments is employed to cooperate with the spectrometer for spectra collection at various temperatures. The lifetime values are obtained from the integral area of the corresponding normalized decay curves.

RESULTS AND DISCUSSION

Structure. The XRD patterns of Yb³⁺ and Nd³⁺-codoped BaY_2O_4 with different concentrations are depicted in Figure 1. Notably, the information from the diffraction peaks of each sample, including relative intensities and locations, is in



Figure 1. XRD patterns of BaY_2O_4 : $x \% Yb^{3+}/y \% Nd^{3+}$ (x = 1, 5, and 10; y = 1, 2, 3, and 5).

accordance with the standard XRD data of BaY_2O_4 (PDF#27-0044), indicating a complete insertion of Yb³⁺ ions and Nd³⁺ ions into the BaY_2O_4 lattice to form a solid solution construction. Moreover, no obvious shift of the diffraction peaks is detected due to the similar ionic radii of Y³⁺ (0.89 Å), Yb³⁺ (0.86 Å), and Nd³⁺ (1.00 Å).

Luminescence Properties. Figure 2a,b displays the emission spectra of the as-prepared samples under the excitation of 980 nm wavelength. Three distinct emission bands appear in the range of 700–900 nm, which are attributed to Nd³⁺: ${}^{4}F_{7/2} \rightarrow {}^{4}I_{9/2}$ at 752 nm, Nd³⁺: ${}^{4}F_{5/2} \rightarrow {}^{4}I_{9/2}$ at 820 nm, and Nd³⁺: ${}^{4}F_{3/2} \rightarrow {}^{4}I_{9/2}$ at 880 nm, respectively. The optimum doping concentrations of both Yb³⁺ and Nd³⁺ in the BaY₂O₄ host are determined to be 1% through a variable-controlling method, by which the strongest Nd³⁺ emission can be acquired. The overall luminescence intensity would be decreased when the doping concentration of Yb³⁺ or Nd³⁺ surpasses the abovementioned value due to the concentration quenching effect.

As shown in Figure 3a, two energy transfer (ET) routes can be involved in the Yb³⁺ and Nd³⁺ codoped system, including a single-photon process (ET1) and a two-photon process (ET2 and ET3). On the one hand, the excited Yb³⁺ can deliver its energy to the Nd³⁺ at the ground state through a phononassisted (PA) ET process, resulting in the population of the Nd^{3+} : ${}^{4}F_{3/2}$ state (ET1). On the other hand, the Nd^{3+} at the ⁴I_{9/2} level can also receive energy directly from the excited Yb³⁺ to jump to the ${}^{4}I_{15/2}$ level (ET2) and then extract energy from another Yb³⁺ to populate to the ${}^{2}H_{11/2}$ level (ET3) followed by an NR process, giving rise to the population of the Nd³⁺: ${}^{4}F_{3/2}$ state. It is important to point out that the excess energy deriving from the whole ET2 and ET3 processes is mainly transferred to heat via phonon vibrations.³⁵ Finally, the Nd³⁺ at the ${}^{4}F_{3/2}$ level can be further excited to the upper ${}^{4}F_{5/2}$ or ${}^{4}F_{7/2}$ level through the assistance of active lattice phonons because of their relatively small ΔE . In order to ulteriorly determine the the ET mechanisms of the Yb^{3+} and Nd^{3+} codoped system, the power density dependence of the emission intensity in BaY_2O_4 : 1% $Yb^{3+}/1\%$ Nd³⁺ is measured and presented in Figure 3b. The relational expression of emission intensity (I)and power density (*P*) is written as $I \propto P^n$. Here, *n* denotes the NIR photon number that needs to be absorbed for emitting a high-energy photon, which can be obtained by the slope of the linear curve of log(I) and log(P). Excited by 980 nm, the n value of the ${}^{4}F_{5/2} \rightarrow {}^{4}I_{9/2}$ transition changes from 0.98 to 1.54 when the excitation power density exceeds 9.69 mW/mm², revealing that the single-photon process is dominant at the low excitation power density, and the two-photon process occurs and gradually increases with the rising excitation power density.

Thermally Enhanced Luminescence. Figure 4a shows the temperature-dependent spectra of BaY_2O_4 : 1% $Yb^{3+}/1\%$ Nd^{3+} under the excitation of 980 nm wavelength, indicating that the emission intensity of the Nd^{3+} : 4F_j (j = 7/2, 5/2, and 3/2) $\rightarrow {}^4I_{9/2}$ transition increases significantly with the rising temperature. This phenomenon can also be observed more intuitively from the digital photos of the far-red luminescence of the sample taken at 298 and 319 K, respectively, which is attributed to the band edge of the Nd^{3+} : ${}^4F_{7/2} \rightarrow {}^4I_{9/2}$ transition (see the inset of Figure 4a). Compared with the room temperature, the intensity of ${}^4F_{7/2} \rightarrow {}^4I_{9/2}$ and ${}^4F_{5/2} \rightarrow {}^4I_{9/2}$ transitions at 548 K increases 180 and 29 times,



Figure 2. UC spectra of (a) BaY_2O_4 : x % $Yb^{3+}/1\%$ Nd³⁺ (x = 1, 5, and 10) and (b) BaY_2O_4 : 10% Yb^{3+}/y % Nd³⁺ (y = 1, 2, 3, and 5).



Figure 3. (a) Possible ET mechanisms in the Yb³⁺ and Nd³⁺ codoped system. (b) Log(I) - log(P) diagram of the ${}^{4}F_{5/2} \rightarrow {}^{4}I_{9/2}$ transition in $BaY_{2}O_{4}$: 1% Yb³⁺/1% Nd³⁺.

respectively, as shown in Figure 4b, which is partly attributed to the efficiency improvement of the PA ET processes between Yb³⁺ and Nd³⁺ benefiting from the intensification of lattice thermal vibration at high temperatures. In addition, the thermal population occurring from the low-lying state to the high-lying state of Nd³⁺ ions should be dramatically enhanced by raising the temperature, giving rise to the depopulation of the corresponding low-lying state. Ultimately, the ${}^{4}F_{7/2}$ level shows the largest thermal enhancement factor, followed by the ${}^{4}F_{5/2}$ level, and the ${}^{4}F_{3/2}$ level with a relatively small trend.

To shed more light on the thermally enhanced luminescence of the Nd³⁺: ⁴F_j (j = 7/2, 5/2, and 3/2) \rightarrow ⁴I_{9/2} transition, the lifetime curves of the Yb³⁺: ²F_{5/2} state in Yb³⁺ single-doped and Yb³⁺/Nd³⁺ codoped samples at various temperatures are collected and the corresponding lifetimes are calculated, as plotted in Figures 5a,b and S1. With the increase in temperature, although the lifetimes of the Yb³⁺: ²F_{5/2} level in BaY₂O₄: 1% Yb³⁺ only show a slight decline, a distinct decrease in the lifetimes of the Yb³⁺: ²F_{5/2} level can be found in the

Yb³⁺/Nd³⁺ codoped sample. Whereafter, the ET rates $(W_{Yb \rightarrow Nd})$ between Yb³⁺ and Nd³⁺ at various temperatures are calculated by the following formula

$$W_{\rm Yb \to Nd} = 1/\tau_{\rm Yb \to Nd} - 1/\tau_{\rm Yb} \tag{1}$$

where $\tau_{Yb \rightarrow Nd}$ and τ_{Yb} are the lifetime values of the Yb³⁺: ${}^{2}F_{5/2}$ level in Yb³⁺/Nd³⁺ codoped and Yb³⁺ singly doped BaY₂O₄, respectively.³⁶ As visualized in Figure 5b, the ET rate between Yb³⁺ and Nd³⁺ significantly increases from 0.15 ms⁻¹ at 298 K to 0.32 ms⁻¹ at 548 K with the increasing temperature, which is a compelling evidence for the thermally promoted ET processes from Yb³⁺ to Nd³⁺ with the assistance of phonons.

Considering the abundant metastable energy levels with small ΔE existing in Nd³⁺ ions, an effective light-to-heat conversion capacity can be anticipated to be obtained by a series of NR processes in BaY₂O₄: Yb³⁺/Nd³⁺ powder. In order to verify this hypothesis, a 980 nm semiconductor laser is used for continuous irradiation on the present sample with an unchanged power density of 13.74 mW/mm², during which



Figure 4. (a) Temperature-dependent spectra of BaY₂O₄: 1% Yb³⁺/1% Nd³⁺ excited by 980 nm wavelength in the range of 298–548 K. The inset shows the luminescence images of the sample at 298 and 319 K taken using a thermal imager. (b) Enhance factor of the Nd³⁺: ${}^{4}F_{j}$ (j = 7/2, 5/2, and 3/2) $\rightarrow {}^{4}I_{9/2}$ transition at various temperatures.



Figure 5. (a) Temperature dependence of lifetime curves of the Yb³⁺: ${}^{2}F_{5/2}$ state in the BaY₂O₄: 1% Yb³⁺/1% Nd³⁺ powder irradiated by 980 nm. (b) Calculated lifetimes of the Yb³⁺: ${}^{2}F_{5/2}$ state in Yb³⁺ single-doped and Yb³⁺/Nd³⁺ codoped samples and the corresponding ET rate between Yb³⁺ and Nd³⁺ at different temperatures. (c) Photothermal conversion effect of BaY₂O₄: 1% Yb³⁺/1% Nd³⁺ as a function of irradiation duration and power density, respectively.

the temperature is monitored by IR thermography. As shown in Figure 5c, the sample temperature is gradually raised with the radiant time extension and comes up to its equilibrium temperature of 319.3 K after approximately 180 s, achieving an increase amplitude of 20.8 K, which is high enough to kill the cancer tissues during the PTT process.³⁷ Subsequently, the power density dependence of the photothermal conversion ability is also conducted with a constant radiant time of 180 s. Distinctly, the temperature of the sample increases monotonously with the increasing power density of 980 nm NIR light. The results mentioned above show the enormous potential of BaY_2O_4 : Yb³⁺/Nd³⁺ powder serving as an extraordinary photothermal conversion agent.

Optical Thermometry Behaviors. In order to discover the optical thermometry properties realized by the thermally coupled Nd³⁺: ⁴F_j (j = 7/2, 5/2, and 3/2) levels, the temperature-dependent luminescence spectra of BaY₂O₄: 1% Yb³⁺/1% Nd³⁺ under the excitation of 980 nm are normalized to 1 at 820 nm, in which the intensity of the ⁴F_{3/2} \rightarrow ⁴I_{9/2} transition is magnified by 10 times for better observation. As shown in Figure 6a, the relative emission intensities of the Nd³⁺: ⁴F_j (j = 7/2, 5/2, and 3/2) \rightarrow ⁴I_{9/2} transition show a regular variation with the increasing temperature, revealing its possibility for optical thermometry. Here, the FIR of ⁴F_{7/2}/⁴F_{5/2}, ⁴F_{5/2}/⁴F_{3/2}, and ⁴F_{7/2}/⁴F_{3/2} are denoted as TCL1, TCL2, and TCL3, respectively, which can be fitted by the Boltzmann distribution law

$$FIR = I_2/I_1 = (\omega_2 A_2 g_2 / \omega_1 A_1 g_1) \cdot \exp(-\Delta E/kT)$$
$$= B \cdot \exp(-\Delta E/kT)$$
(2)



Figure 6. Temperature dependence of (a) emission spectra, (b) absolute sensitivity S_A , and (c) relative sensitivity S_R of BaY₂O₄: 1% Yb³⁺/1% Nd³⁺ excited by 980 nm wavelength.

where I, A, ω , g, T, k, and B represent the emission intensity, spontaneous emission rate, angular frequency, degeneracy, absolute temperature, Boltzmann's constant, and a temperature-independent constant, respectively.^{38–43} The corresponding FIR fitting curves are presented in Figure S2, from which the ΔE values are calculated to be 858 cm⁻¹ (${}^{4}F_{7/2}/{}^{4}F_{5/2}$), 1252 cm⁻¹ (${}^{4}F_{5/2}/{}^{4}F_{3/2}$), and 2022 cm⁻¹ (${}^{4}F_{7/2}/{}^{4}F_{3/2}$) for TCL1, TCL2, and TCL3 respectively.

As the important parameters of the optical thermometer, the fitting curves of the absolute sensitivity S_A and relative sensitivity S_R are outlined by the following equations

$$S_{\rm A} = |{\rm d}({\rm FIR})/{\rm d}T| = {\rm FIR} \cdot \Delta E/(k_{\rm B} \cdot T^2)$$
(3)

$$S_{\rm R} = |{\rm d}({\rm FIR})/({\rm FIR}) \cdot {\rm d}T| = \Delta E/(k_{\rm B} \cdot T^2)$$
⁽⁴⁾

As shown in Figure 6b, the S_A values of TCL1, TCL2, and TCL3 increased with the rising temperature and reached their corresponding maximal values of 0.15, 68.88⁻⁻, and 40.13% K⁻¹ at 548 K, respectively. On the contrary, the S_R values of TCL1, TCL2, and TCL3 decreased with the increasing

temperature and reached their corresponding maximum values of 1.39, 2.03, and 3.29% K^{-1} at the initial temperature, respectively, as illustrated in Figure 6c. Utilizing the obtained $S_{\rm A}$ and $S_{\rm R}$ values, we can quantitatively compare the temperature sensing performance of BaY2O4: 1% Yb3+/1% Nd³⁺ with others. Table 1 enumerates a series of temperature sensing materials based on the thermally coupled Nd³⁺: ${}^{4}F_{i}$ (*j* = 7/2, 5/2, and 3/2) level, from which it can be seen that the present sample owns the highest absolute sensitivity S_A and its relative sensitivity S_R is only slightly less than that of oxygen fluorine glass, indicating the excellent optical thermometry ability of BaY₂O₄: Yb³⁺/Nd³⁺. Moreover, the accuracy of the present optical thermometer is evaluated by a self-designed experiment in which a heat gun is employed to heat the sample, and its temperature is determined by an IR temperature monitor and the FIR technology. As shown in Figure S3, the temperature calculated by FIR technology matches well with that detected using the IR thermometer, especially TCL3, with a maximum absolute error of 3 K. In addition, the FIR values of TCL1, TCL2, and TCL3 remained almost unchanged after conducting five cyclic processes, revealing their outstanding reproducibility for optical thermometry, as depicted in Figure S4.

Ex Vivo Examinations in Biotissues. A simple ex vivo test is devised to estimate the penetration depth of the Nd³⁺: ${}^{4}F_{j}$ (j = 7/2, 5/2, and 3/2) $\rightarrow {}^{4}I_{9/2}$ transition in BaY₂O₄: 1% Yb³⁺/1% Nd³⁺, as illustrated in Figure 7a. The collected spectra shown in Figure 7b are all measured at the ambient temperature. Distinctly, the far-red and NIR luminescence of Nd³⁺ is gradually decreased with the thickened chicken muscle tissues. However, the spectral profile remains almost unchanged with the adding thickness of the biological tissues. Moreover, its spectral information can still be gathered under 7 mm-thick biotissue due to the weak absorption and scattering effect of the biological tissues in this region, portending its potential application in the deep tissues.

CONCLUSIONS

In summary, a newfashioned photothermal conversion agent BaY_2O_4 : Yb^{3+}/Nd^{3+} integrating with real-time temperature

Table 1. Various Nd³⁺-Based Optical Thermometers with Related Parameters

materials	dopant	EX	TCLs	wavelength (nm)	T(K)	$\Delta E \ (cm^{-1})$	$S_{\rm R} ({\rm K}^{-1})$	$S_{\rm A} \ (\% \ {\rm K}^{-1})$	ref.
BaY_2O_4	Yb ³⁺ /Nd ³⁺	980 nm	${}^{4}F_{7/2}$, ${}^{4}F_{5/2}$	700-900	298-548	858	$1234/T^2$	0.15	this work
			${}^{4}F_{5/2}$, ${}^{4}F_{3/2}$	700-900	298-548	1252	$1800/T^{2}$	68.88	this work
			${}^{4}F_{7/2}$, ${}^{4}F_{3/2}$	700-900	298-548	2022	$2908/T^2$	40.13	this work
LaPO ₄	Yb^{3+}/Nd^{3+}	980 nm	${}^{4}F_{7/2}$ ${}^{4}F_{5/2}$	700-900	280-490	938	$1349/T^2$	0.11	21
			${}^{4}F_{5/2}$ ${}^{4}F_{3/2}$	700-900	280-490	935	$1345/T^2$	18.53	21
			${}^{4}F_{7/2}$, ${}^{4}F_{3/2}$	700-900	280-490	1897	$2727/T^{2}$	7.36	21
oxyfluoride glass	Yb^{3+}/Nd^{3+}	980 nm	⁴ F _{7/2} , ⁴ F _{5/2}	720-900	303-623	1300	$1763/T^{2}$		36
			${}^{4}F_{5/2}$, ${}^{4}F_{3/2}$	720-900	303-623	1216	$1884/T^{2}$		36
			${}^{4}F_{7/2}$, ${}^{4}F_{3/2}$	720-900	303-623	2076	$3010/T^2$		36
La_2O_3	Yb^{3+}/Nd^{3+}	980 nm	${}^{4}F_{7/2}$, ${}^{4}F_{5/2}$	725-850	293-1233	927	$1344/T^{2}$	0.03	38
CaWO ₄	Yb ³⁺ /Nd ³⁺ /Li ⁺	980 nm	${}^{4}F_{7/2}$, ${}^{4}F_{5/2}$	700-950	303-773	839	$1207/T^{2}$	0.01	44
			${}^{4}F_{5/2}$, ${}^{4}F_{3/2}$	700-950	303-773	865	$1244/T^2$	0.39	44
			${}^{4}F_{7/2}$, ${}^{4}F_{3/2}$	700-950	303-773	1722	$2476/T^2$	0.36	44
$Bi_4Ti_3O_{12}$	Yb ³⁺ /Nd ³⁺	980 nm	${}^{4}F_{7/2}$, ${}^{4}F_{5/2}$	700-850	323-573	867.2	$1248/T^2$	9.9	45
			${}^{4}F_{7/2}$, ${}^{4}F_{5/2}$	720-950	297-420	889	$1288/T^{2}$		46
NaYF ₄	Yb ³⁺ /Nd ³⁺	980 nm	${}^{4}F_{5/2}$, ${}^{4}F_{3/2}$	720-950	297-420	713	$1033/T^{2}$		46
			${}^{4}F_{7/2}, {}^{4}F_{3/2}$	720-950	297-420	1433	$2076/T^2$		46



Figure 7. (a) Diagrammatic sketch of the ex vivo examination conducted at room temperature. (b) Biotissue thickness-dependent luminescence spectra of BaY_2O_4 : 1% $Yb^{3+}/1\%$ Nd³⁺ under the excitation of 980 nm along with the corresponding integral intensity.

sensing has been designed successfully, which can be used for deep-tissue treatment because of the excitation and emission wavelengths located in the biological window. The optical thermometry behavior is based on the FIR of the thermally coupled Nd³⁺: ${}^{4}F_{j}$ (*j* = 7/2, 5/2, and 3/2) level with the maximum S_A and S_R of 68.88 and 3.29% K⁻¹, respectively. More importantly, the emission intensity of the Nd³⁺ ion is significantly enhanced by increasing the temperature due to the improvement of PA ET efficiency, allowing the acquisition of a better signal-to-noise ratio during the temperature detection process. Meanwhile, the present sample shows remarkable photothermal conversion performance under the irradiation of 980 nm. The sample temperature can be raised to 319.3 K after 180 s of NIR irradiation with a fixed power density of 13.74 mW/mm², which is effective enough to destroy the pathological tissues. In addition, the penetration depth of the prepared sample in the biotissues was determined to be approximately 7 mm through a simple ex vivo examination. The abovementioned experimental data indicate that BaY₂O₄: Yb³⁺/Nd³⁺ is a promising photothermal conversion agent equipped with an accurate temperature sensing function, which can be utilized in the deep-tissue PTT process.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.inorgchem.2c00432.

Temperature dependence of decay curves of the Yb³⁺: ${}^{2}F_{5/2}$ level, temperature-dependent FIR curves, temperature of the sample obtained using an IR thermometer and FIR technology, and repeatability studies of TCL1, TCL2, and TCL3 (PDF)

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Notes

The authors declare no competing financial interest.

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