



Modified band alignment at multilayer MoS₂/Al₂O₃ heterojunctions by nitridation treatment

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ABSTRACT

Interface engineering plays a decisive role in the two-dimensional material/high- κ oxide heterojunction-based devices. In this work, the effects of NH₃ plasma interface treatment on the energy band alignment at multilayer MoS₂/Al₂O₃ heterostructures were explored using x-ray photoelectron spectroscopy. A type-I band alignment was formed with valence band offset (VBO) and conduction band offset (CBO) about 3.67 eV and 2.33 eV, respectively. Significantly, the CBO was enlarged by about 0.45 eV after the plasma treatment, which was mainly attributed to the increased separation between the Al 2p core-level energy and valence band maximum. This interesting finding provides a significant guidance for the band adjustment at the interface and further applications of multilayer MoS₂ in electronic devices.

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1. Introduction

In modern semiconductor-based devices such as lasers, light emitting diodes, photodetectors, transistors and so on, heterojunctions (HJs) play a key role owing to that their electrical and optical properties are mainly controlled by the band alignment at the HJs. However, the lattice mismatch and interfacial defects at the interfaces severely limit the development of HJ devices based on conventional bulk semiconductors. Fortunately, since the discovery of graphene in 2004 [1], atomically thin two dimensional (2D) materials have opened a new door for the development of semiconductor devices due to their unique structures, electronic, optical, mechanical, and thermal properties [2–4]. Especially, the formation of van der Waals bonded heterostructures between 2D layers potentially permit to create any desirable HJs [5–7]. Among

the 2D semiconductors, MoS₂ as a typical representative has drawn an increasing attention due to its wide applications in optoelectronic and nanoelectronic technologies [8–11]. Monolayer MoS₂ contains a hexagonal trilayer with one Mo layer sandwiched between two sulfur layers, and the neighboring MoS₂ layers were weakly attached by van der Waals forces. Meanwhile, the band gap of MoS₂ was strongly related to the layer number, with a direct bandgap of about 1.80 eV for monolayer and indirect bandgap of 1.20 eV for bulk layers. Although multilayer MoS₂ have been seriously restricted in the application of photonic devices by the indirect bandgap, it has already been shown to enable the channel of metal-insulator-semiconductor transistors scaled down profoundly [12–14]. Furthermore, multilayer MoS₂ can facilitate transistors with very steep sub-threshold slope and high drive current which is attribute to the high density of states [15,16].

Currently, multilayer MoS₂/wide bandgap insulator HJs have been an indispensable part in the MoS₂-based electronic devices such as high frequency integrated logic circuits and nonvolatile memory. SiO₂ as the typical and most common used oxide insulator has been unable to satisfy the requirement of enhanced specific capacitance, low gate leakage current, and high carrier mobility in

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the related devices. High- κ oxides (e.g. Al_2O_3) were proved to improve the carrier mobility significantly owing to the dielectric screening effects [17,18]. Therefore, high- κ oxides have great potential applications in multilayer-based MoS_2 transistors. In addition, it is essential to point out that the characteristics of the MoS_2 -based electronic devices are critically sensitive to the band alignment at the MoS_2 /oxide HJ interface which strongly affect electron and hole transport property in the form of conduction band offset (CBO) and valence band offset (VBO). The VBO and CBO should be sufficiently large enough to reduce the injection of holes and electrons [19], respectively. Meanwhile, the defects and charges located at the semiconductor/oxide HJ interface also exert great influences on the energy band profile and need to be optimized using passivation technology. For Al_2O_3 oxide, NH_3 treatment on the surface has already been proved to be an effective surface engineering method to reduce the surface or interface defects [20]. In this paper, we investigated the band alignment at multilayer MoS_2 / Al_2O_3 HJs, and the effects of NH_3 treatment on the band offset at MoS_2 / Al_2O_3 interface was explored by x-ray photoelectron spectroscopy (XPS) measurements, which provides a significant guidance for the band adjustment at the interface and further applications of MoS_2 in electronic devices.

2. Experiments

The multilayer MoS_2 / Al_2O_3 samples investigated in this paper were prepared by the PMMA method. Firstly, large area, continuous multilayer MoS_2 films were grown on SiO_2 /Si substrates using chemical vapor deposition method as shown in Fig. 1(a). During the growing process, MoO_3 and sulfur powder were employed as the Mo and S sources with Ar as the carrier gas, respectively. Meanwhile, the growth temperature was located at 800°C , and the growth time was about 5 min. Then, the Al_2O_3 dielectric oxide on Si was grown by atomic layer deposition system at 200°C , using trimethylaluminum and H_2O as precursors, and the thickness was about 8.0 nm. The nitridation process on Al_2O_3 surface was carried out by employing low-energy NH_3 plasma treatments. By optimizing the conditions of plasma treatment in order to reduce the defect density at the interface, the RF power and flow rate were fixed at 200 W and 150 sccm, respectively. Meanwhile, the plasma treatment time was about 120 s and the pressure was kept at 100 Pa during the process. Then obvious N 1s peak located at about 398.74 eV was observed in the XPS spectrum shown in Fig. 1(b). For the reference sample, no NH_3 plasma treatment was implemented on Al_2O_3 surface. Lastly, the multilayer MoS_2 film were transferred onto the prepared Al_2O_3 /Si substrates. Raman spectra at room temperature were collected through a Raman spectrophotometer system (Horiba Scientific, Labram HR evolution). The XPS measurements were performed using a VG ESCALAB 220i-XL system with a monochromatic Al K α source at the power of 150 W. The photon energy of the x-ray source is about 1486.6 eV. During the measurements, XPS spectra were acquired using 20 eV pass energy, and a charge neutralizer was used to prevent charging. The XPS binding energy was calibrated using pure gold, silver, and copper standard samples by setting the Au 4f $_{7/2}$, Ag 3d $_{5/2}$, and Cu 2p $_{3/2}$ peaks at 83.986 ± 0.02 eV, 368.266 ± 0.02 eV, and 932.676 ± 0.02 eV, respectively. The Fermi level edge was calibrated using pure nickel (Ni) and setting the binding energy at 0.00 ± 0.02 eV. In addition, C 1s peak (284.8 eV) was used to correct the core-level binding energy in order to eliminate the sample surface differential charging effect.

3. Results and discussions

Raman spectroscopy as a powerful tool to investigate the optical

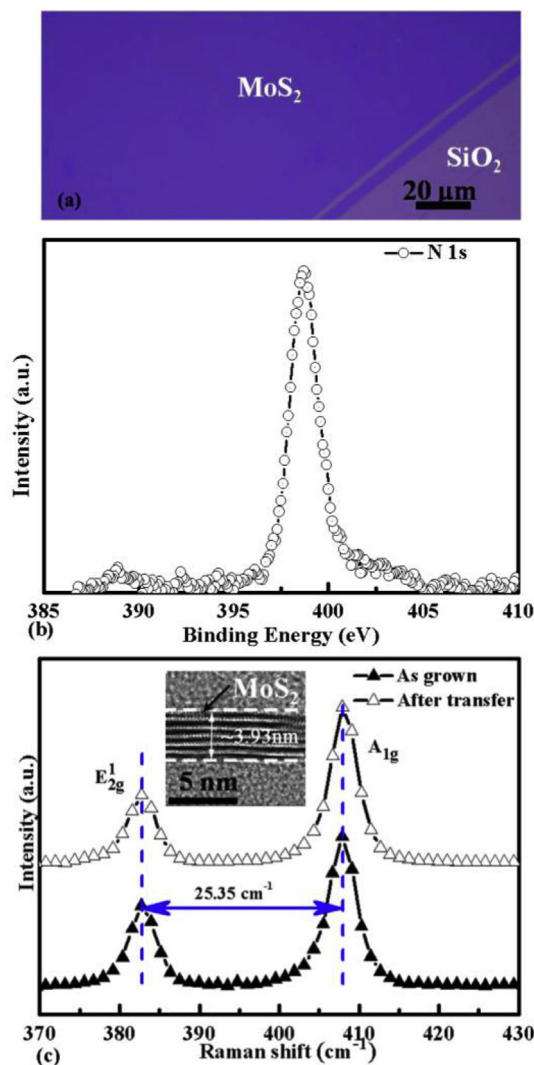


Fig. 1. (a) The optical micrograph of large area multilayer MoS_2 grown on SiO_2 /Si substrate; (b) the N 1s XPS spectrum of the Al_2O_3 with NH_3 plasma treatment; (c) The Raman spectroscopy of the multilayer MoS_2 film before and after transfer. The inset is the cross-sectional TEM image of the MoS_2 , which identified the layers of MoS_2 .

quality and to ascertain the layer number of 2D materials was performed for the samples. Fig. 1(c) shows the Raman spectrum of the multilayer MoS_2 which were grown on SiO_2 /Si substrate and transferred onto Al_2O_3 oxide. Obviously, two typical Raman modes, E_{2g}^1 (382.70 cm^{-1}) and A_{1g} (408.05 cm^{-1}) were observed, and the difference between them is 25.35 cm^{-1} , indicating the film layers was more than five [21]. Furthermore, the cross-sectional transmission electron microscopy (TEM) result displayed in the inset of Fig. 1(c) confirmed this inference as shown that the thickness is about 3.93 nm, corresponding to six layers. Meanwhile, the Raman position and full width at the half maximum of the peaks almost kept the same before and after transfer, demonstrating the optimized transfer process largely retained the material quality.

XPS has been extensively employed to investigate the energy band offset at MoS_2 -based HJs, such as MoS_2 /GaN [22], MoS_2 / $\text{In}_{0.15}\text{Al}_{0.85}\text{N}$ [23], MoS_2 / SiO_2 [24], MoS_2 / HfO_2 [25] and so on. The energy core level and valence band spectra of MoS_2 and Al_2O_3 were shown in Fig. 2. The VBO ΔE_v can be calculated according the following formula [26].

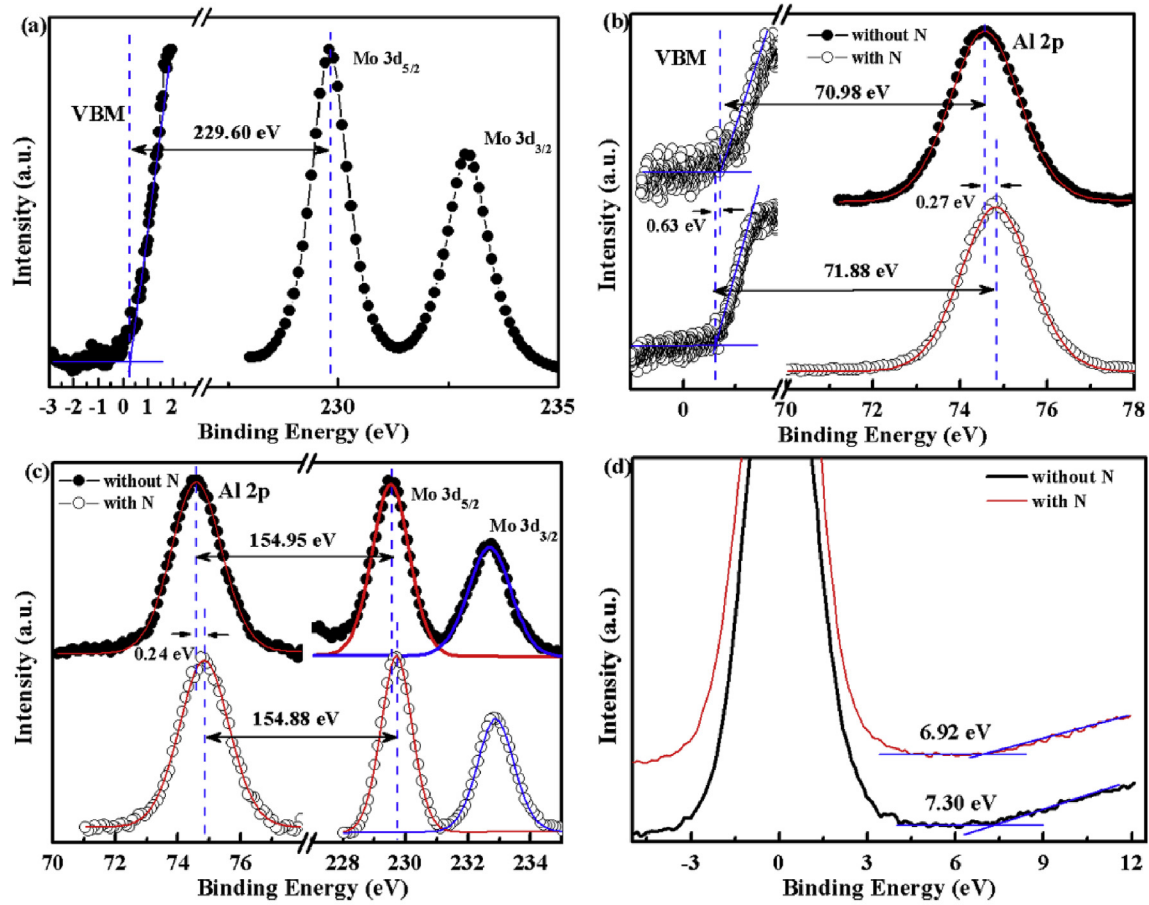


Fig. 2. The core-level and valence band spectra for (a) multilayer MoS₂ and (b) bare Al₂O₃ without and with NH₃ plasma treatment. (c) The corresponding Al 2p and Mo 3d core-level energy spectrum collected from both multilayer MoS₂/Al₂O₃ samples. The VBM is obtained from the intercept of the slope at the leading edge of the valence band spectrum with the base line. (d) The O 1s photoelectron energy loss spectra for the Al₂O₃ oxide.

$$\Delta E_v = (E_{cl}^{MoS_2} - E_v^{MoS_2}) - (E_{cl}^{Al_2O_3} - E_v^{Al_2O_3}) - (E_{cl}^{H-MoS_2} - E_{cl}^{H-Al_2O_3}) \quad (1)$$

where $E_{cl}^{MoS_2}$ and $E_v^{MoS_2}$ represent the core-level and valence band maximum (VBM) energy value of MoS₂ obtained from multilayer MoS₂, respectively; $E_{cl}^{Al_2O_3}$ and $E_v^{Al_2O_3}$ are the related core-level and VBM energy of bare Al₂O₃, respectively; $E_{cl}^{H-MoS_2}$ and $E_{cl}^{H-Al_2O_3}$ correspond to the core-level energy of MoS₂ and Al₂O₃, respectively, which were derived from the multilayer MoS₂/Al₂O₃ HJs. Fig. 2(a) showed the Mo 3d core-level and valence band spectrum collected from multilayer MoS₂. Apparently, the Mo 3d core level was reconstituted with 3d_{5/2} (229.86 ± 0.05 eV) and 3d_{3/2} (232.94 ± 0.05 eV) two peaks, and the binding energy difference was about 3.08 eV induced by the spin-orbit splitting. The VBM with respect to the Fermi level energy (located at zero) was extrapolated from the intercept between the slope of the leading edge and the base line [27,28]. Then the energy separation between Mo 3d_{5/2} and VBM was 229.60 ± 0.05 eV. As a result, the first bracket term in Eq. (1) is about 229.60 eV just as labeled in Fig. 2(a). For the bare Al₂O₃ oxide without NH₃ treatment, the VBM was 3.59 ± 0.05 eV, while the surface nitridation made the VBM shifted towards Fermi level by 0.63 ± 0.05 eV as shown in Fig. 2(b), indicating this passivation technology exerted a significant influence on the valence band of Al₂O₃. Similarly, the core-level binding energy

of Al 2p was shifted upwards 0.27 ± 0.05 eV after NH₃ treatment in comparison with that without nitridation. Then, the second bracket term in Eq. (1) was 70.98 ± 0.05 eV and 71.88 ± 0.05 eV for the multilayer MoS₂/Al₂O₃ samples without and with NH₃ plasma treatment, respectively. Furthermore, Fig. 2(c) showed the core-level binding energy separation between Al 2p and Mo 3d_{5/2} obtained from multilayer-MoS₂/Al₂O₃ HJs, 154.95 ± 0.05 eV for the sample without interface passivation while 154.88 ± 0.05 eV for the plasma treated sample, which corresponded to the third bracket term values in Eq. (1). Consequently, the ΔE_v between multilayer MoS₂ and Al₂O₃ was about 3.67 ± 0.05 eV and 2.84 ± 0.05 eV for the sample without and with NH₃ treatment, respectively. Subsequently, the CBO value ΔE_c was calculated based on the energy band gap of Al₂O₃ ($E_g^{Al_2O_3}$) and multilayer MoS₂ ($E_g^{MoS_2}$) expressed as

$$\Delta E_c = E_g^{Al_2O_3} - \Delta E_v - E_g^{MoS_2} \quad (2)$$

where the $E_g^{Al_2O_3}$ was obtained from the O1s loss energy spectrum as shown in Fig. 2(d). Obviously, $E_g^{Al_2O_3}$ for the Al₂O₃ without NH₃ treatment was about 7.30 ± 0.05 eV extrapolated from the separation between the core-level energy of Al–O bonds and the linear edge base line. In compare with this sample, interface nitridation reduced the energy band gap of Al₂O₃ by about 0.38 eV. While for multilayer MoS₂, the electronic band gap $E_g^{MoS_2}$ was set at 1.30 eV [29] in this work. As a result, the ΔE_c value for the samples without and with plasma treatment determined from Eq. (2) was

by nitridation in comparison with the sample without NH_3 treatment. Meanwhile, the interface mid-gap states may enhance the electron transfer between the MoS_2 and Al_2O_3 , which contributes to the upshift of Mo 3d core-level energy. For MoS_2 -based electronic devices, the mid-gap states may deteriorate the device performances acting as electron leakage channel. However, herein, for multilayer- MoS_2 FETs, the electrons were mainly transported in the top few layers due to the electrons were scattered away from the interface by the enhanced potential barrier of the conduction band offset. For monolayer- $\text{MoS}_2/\text{Al}_2\text{O}_3$ heterojunctions, this effect needs to be further investigated and optimized. Therefore, the NH_3 plasma treatment is beneficial to the multilayer MoS_2 based electronic devices.

4. Conclusions

In conclusion, the influences of NH_3 plasma treatment to Al_2O_3 surface on the band alignment of multilayer- $\text{MoS}_2/\text{Al}_2\text{O}_3$ heterostructures were explored experimentally and theoretically. The formation of the nitridation interfacial layer significantly enlarge the conduction band offset by about 0.45 eV, which was primarily attributed to the increased separation between the Al 2p core-level and valence band maximum. The surface passivation by NH_3 plasma treatment provides a promising way to modulate the energy band alignment at heterointerfaces, which exerts great impacts on the related devices.

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