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ORIGINAL ARTICLE

Observation of efficient population of the red-emitting state from the green state by non-multiphonon relaxation in the $Er^{3+}-Yb^{3+}$ system

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The rare earth Er^{3+} and Yb^{3+} codoped system is the most attractive for showcasing energy transfer upconversion. This system can generate green and red emissions from Er^{3+} under infrared excitation of the sensitizer Yb^{3+} . It is well known that the red-emitting state can be populated from the upper green-emitting state. The contribution of multiphonon relaxation to this population is generally considered important at low excitation densities. Here, we demonstrate for the first time the importance of a previously proposed but neglected mechanism described as a cross relaxation energy transfer from Er^{3+} to Yb^{3+} , followed by an energy back transfer within the same $Er^{3+}-Yb^{3+}$ pair. A luminescence spectroscopy study of cubic $Y_2O_3:Er^{3+}$, Yb^{3+} indicates that this mechanism can be more efficient than multiphonon relaxation, and it can even make a major contribution to the red upconversion. The study also revealed that the energy transfers involved in this mechanism take place only in the nearest $Er^{3+}-Yb^{3+}$ pairs, and thus, it is fast and efficient at low excitation densities. Our results enable a better understanding of upconversion processes and properties in the $Er^{3+}-Yb^{3+}$ system. *Light: Science & Applications* (2015) **4**, e239; doi:10.1038/lsa.2015.12; published online 16 January 2015

Keywords: energy transfer; erbium-ytterbium system; upconversion luminescence

INTRODUCTION

Infrared to visible upconversion luminescence has been extensively studied for its fundamental value^{1–3} and its various potential applications in upconversion lasers,⁴ bioimaging,⁵ *etc.* The codoping of Er³⁺ and a high concentration of sensitizer Yb³⁺ forms the most attractive energy transfer upconversion (ETU) system. Under infrared (980 nm) excitation of the sensitizer Yb³⁺, this system can generate green and red upconversions originating from the ${}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2}$ and ${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$ transitions of Er³⁺, respectively. Unlike the green upconversion, the red upconversion benefits from several possible excitation mechanisms.^{6,7} Multiphonon relaxation (MPR) from the upper ${}^{4}S_{3/2}$ state and ETU from the lower intermediate ${}^{4}I_{13/2}$ state are generally considered dominant at low infrared excitation densities because other mechanisms involving three photon processes^{6,7} become important only at high infrared excitation densities,⁸ which is not the topic of this work.

The MPR is not the only mechanism for populating the ${}^{4}F_{9/2}$ from the ${}^{4}S_{3/2}$; a non-MPR mechanism was proposed earlier,⁸ but it has not been considered important since then. This mechanism involves two sequential energy transfers between Er^{3+} and Yb^{3+} . The first step is a well-known cross-relaxation (CR) energy transfer from Er^{3+} in the ${}^{4}S_{3/2}$ excited state (${}^{4}S_{3/2} \rightarrow {}^{4}I_{13/2}$) to Yb^{3+} in the ground state (${}^{2}F_{5/2} \leftarrow {}^{2}F_{7/2}$),⁹ resulting in the excitation of $Er^{3+} {}^{4}I_{13/2}$ and $Yb^{3+} {}^{2}F_{5/2}$. The subsequent step is an energy back transfer from the $Yb^{3+} ({}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2})$ excited by the CR to Er^{3+} in the ${}^{4}I_{13/2}$ state (${}^{4}F_{9/2} \leftarrow {}^{4}I_{13/2}$) to promote the excitation of $Er^{3+} {}^{4}F_{9/2}$. Hence, the CR can be divided

into two parts: one is followed by the energy back transfer, CRB, and the other is not followed by the energy back transfer, CRNB. The Yb³⁺- induced green emission quenching of Er^{3+} by the CR has been widely recognized.^{9,10} However, the population of the ${}^{4}\text{F}_{9/2}$ by the CRB from the ${}^{4}\text{S}_{3/2}$ has rarely been studied or valued in both photoluminescence (PL) and upconversion luminescence (UCL).

In this article, we report an observation of the CRB in cubic Y_2O_3 : Er^{3+} , Yb^{3+} . We find the CRB can be more efficient than MPR and can even make a major contribution to the red UCL. To the best of our knowledge, this is the first time that the CRB has been found to be important to the red emission both in the PL and UCL of the Er^{3+} – Yb^{3+} system.

MATERIALS AND METHODS

Sample preparation

The cubic $Y_2O_3:0.002Er^{3+}$, xYb^{3+} (x=0, 0.04, 0.1, 0.2, 0.3) samples were prepared by the normal firing precursor method.² The firing precursor is more favorable to achieve uniform and highly crystallized samples than the traditional solid-state reaction. The starting aqueous solutions, $Y(NO_3)_3$, $Yb(NO_3)_3$, and $Er(NO_3)_3$, with corresponding mole ratios were mixed and stirred vigorously to form a homogeneous solution. The samples were obtained after being dried at 100 °C for 6 h and then calcined at 1600 °C for 6 h. The expression of $Y_2O_3:0.002Er^{3+}$, xYb^{3+} in the present paper means the formula $Y_{2-0.002-x}Er_{0.002}Yb_xO_3$. The low Er^{3+} and high Yb^{3+} concentrations

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Received 4 August 2014; revised 12 November 2014; accepted 18 November 2014; accepted article preview online 19 November 2014

applied in this work were used to achieve a real ETU system and suppress the interaction among Er^{3+} ions for simplifying the ETU processes.

Spectroscopy measurements

Steady state PL and UCL spectra were measured using an EI-FS920 fluorimeter with a xenon lamp as an excitation source for PL and a CW 980 nm laser diode as an excitation source for UCL. The decay curves of PL and UCL were detected using a Triax 550 spectrometer (Jobin-Yvon) and recorded by a Tek-tronix digital oscilloscope (TDS 3052), while a 10 ns pulsed laser with tunable wavelengths from an optical parametric oscillator pumped by a Nd:YAG laser (spectra-physics, GCR 130) was used as an excitation source. In energy level lifetime measurements, the excitation wavelengths were tuned to 520 nm for $Er^{3+} {}^{4}S_{3/2}$, 650 nm for $Er^{3+} {}^{4}F_{9/2}$ and ${}^{4}I_{11/2}$, 1480 nm for $Er^{3+} {}^{4}I_{13/2}$ and 980 nm for Yb³⁺ ${}^{2}F_{5/2}$. The lifetime is defined as the area under the decay curve with normalized initial intensity. The UCL spectra under pulse excitation at 980 nm were detected using a USB4000 spectrometer (Ocean Optics), which gives the time-integrated intensities.

RESULTS AND DISCUSSION

Observation of the CRB in PL

Figure 1 shows PL spectra of Y_2O_3 :0.002 Er^{3+} , xYb^{3+} under 520 nm excitation of $\text{Er}^{3+}{}^2\text{H}_{11/2}$, which can rapidly relax to the ${}^4\text{S}_{3/2}$ due to their proximity in energy. One can observe that the red (660 nm) to green (560 nm) emission intensity ratio (R/G) increases considerably with increasing *x*. A similar result upon $\text{Er}^{3+}{}^4\text{F}_{7/2}$ excitation with 488 nm was also observed in Y_2O_3 : Er^{3+} , Yb^{3+} nanocrystals,¹¹ and an interaction between two excited Er^{3+} ions (${}^4\text{F}_{7/2} \rightarrow {}^4\text{F}_{9/2}$ and ${}^4\text{F}_{9/2} \leftarrow {}^4\text{I}_{11/2}$)⁷ was considered the main excitation mechanism of the red-emitting ${}^4\text{F}_{9/2}$ state. In the present work, the ${}^4\text{F}_{7/2}$ is not excited and the Er^{3+} concentration is low, so the above mechanism is not feasible. We found through experimentation that the R/G ratio remains unchanged if La^{3+} or Lu^{3+} is doped instead of Yb^{3+} , indicating that the ${}^4\text{S}_{3/2}-{}^4\text{F}_{9/2}$ MPR rate hardly changes with rare earth doping. Therefore, we attribute the Yb^{3+} -induced R/G increase to the effect of CRB, which benefits from well-matched energy level structures between Er^{3+} and Yb^{3+} , as sketched in the insert to Figure 1. The



Figure 1 PL spectra upon ${}^{2}H_{11/2}$ excitation at 520 nm. The green emission intensities are normalized, and the red ones (620–700 nm) are magnified 10 times. The insert shows the energy level diagram with the CR and CRB processes. CR, cross-relaxation; PL, photoluminescence.

CRB may take place within the same Er³⁺-Yb³⁺ pair, meaning that the Yb^{3+} ion excited by the CR transfer from Er^{3+} transfers its energy back again to the same Er³⁺ ion. The CRB rate in this case is independent of excitation densities, similar to the MPR. If the CRB takes place not within the same $Er^{3+}-Yb^{3+}$ pair, i.e., the excitation energy of Yb^{3+} is not retransferred to the same Er^{3+} but transferred to any other Er^{3+} ion in the ${}^{4}I_{13/2}$ state, then the CRB rate in this case is dependent on the concentration of the other Er³⁺ ions and thus dependent on excitation densities. Therefore, the CRB may become important only at high excitation densities. In the present work, the Yb³⁺-induced R/G increase can be observed under weak 520 nm excitation using a grating monochromator with a xenon lamp. We found that the enlarged R/G ratio in the presence of Yb³⁺ is independent of excitation densities using the xenon lamp as an excitation source. Based on the CRB models mentioned above, we consider that the CRB takes place within the same $Er^{3+}-Yb^{3+}$ pair in the samples in this work.

Time evolutions of PL are measured for distinguishing the CRB from the MPR, as shown in Figure 2. With increasing x, the appreciable shortening of the ${}^{4}S_{3/2}$ lifetime (τ_{4}) (Figure 2a) is due to the CR,^{9,10} and the small shortening of the ${}^{4}F_{9/2}$ lifetime (τ_{3}) (Figure 2b) is due to weak coupling¹² of $\text{Er}^{3+4}\text{F}_{9/2} \rightarrow 4\text{I}_{15/2}$ with $\text{Yb}^{3+2}\text{F}_{5/2} \leftarrow 2\text{F}_{7/2}$. In the time evolution of the red emission after the ${}^{4}S_{3/2}$ is populated (Figure 2c), we find that the samples containing Yb^{3+} appear to undergo a fast build-up process, unlike the Yb³⁺-free sample that appears as a normal slow rising edge.¹³ Accordingly, the time evolu-tion patterns in the presence of Yb³⁺ shown in Figure 2c are the combined patterns of the fast build-up component and the slow build-up component. If one normalizes the initial emission intensity of the fast build-up component (Figure 2c) and denotes the total area under the time evolution curve by T_3 , the proportion of the emitted red photons by the fast build-up component of the total emitted red photons can be calculated by τ_3/T_3 . This proportion is found to be consistent with the proportion of the increment in the R/G ratio (Table 1), written as $\alpha = 1 - (\tau_3/\tau_{3,0})(R/G)_0/(R/G)$, where the subscript 0 identifies the case of x=0 and $(\tau_3/\tau_{3,0})(R/G)_0$ is the R/G ratio contributed only by MPR in the presence of Yb³⁺. The result indicates that



Figure 2 Time evolutions of the green and red emissions.

Table 1 PL data used to identify the CRB process in $Y_2O_3{:}0.002 Er^{3+}, \, {\it xYb}^{3+}$

x	R/G	<i>I</i> ₄	τ ₄ (μs)	τ ₃ (μs)	τ_{3}/T_{3}	α
0	1	1	124	24	0	0
0.04	1.4	0.66	86	22.5	0.29	0.33
0.10	2.15	0.33	66	20.9	0.54	0.59
0.20	3.5	0.19	43	18.9	0.73	0.78
0.30	5.65	0.1	30	17.2	0.84	0.87

Abbreviation: PL, photoluminescence.

the CRB is a fast process. Accordingly, there must be a fast CR. However, a corresponding fast decay of the green emission is not detected (Figure 2a). We thus deduce that the fast green emission may be completely quenched by the fast CR. This deduction is further evidenced in the following section.

Efficiency of the CRB in PL

After the ${}^{4}S_{3/2}$ is populated upon 520 nm excitation, the ${}^{4}S_{3/2}$ undergoes depopulation by radiative transition, MPR, CRNB and CRB, with respective efficiencies of η_{r4} , η_{MPR43} , η_{CRNB} and η_{CRB} . Figure 3 shows the comparison of PL spectra under 520 nm excitation with that under 650 nm excitation of the ${}^{4}F_{9/2}$. Here, the red intensity under 650 nm excitation. Under 520 nm excitation, one can find that the green intensity enhance ment of $Er^{3+4}I_{13/2} \rightarrow {}^{4}I_{15/2}$ emission at 1530 nm and $Yb^{3+2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ emission at 1040 nm compared to that under 650 nm excitation. The feature is an explicit indication of the CRNB process. In the absence of Yb^{3+} , the $Er^{3+4}I_{11/2} \rightarrow {}^{4}I_{15/2}$ emissions for both 520 nm and 650 nm



Figure 3 Comparison of PL spectra upon $^2\text{H}_{11/2}$ excitation at 520 nm with that upon $^4\text{F}_{9/2}$ excitation at 650 nm. The red emission intensity under 650 nm excitation is normalized to that under 520 nm excitation. The red emission intensities (600–700 nm) are magnified 20 times. PL, photoluminescence.

excitations can be observed at 1000 nm; they coincide because the population of ${}^{4}I_{11/2}$ by ${}^{4}S_{3/2} \rightarrow {}^{4}I_{9/2}$, ${}^{4}I_{11/2}$ radiative transitions for 520 nm excitation is very small with respect to that from the ${}^{4}F_{9/2}$ by cascade MPR. In the presence of Yb³⁺, the Er³⁺ ${}^{4}I_{11/2} \rightarrow {}^{4}I_{15/2}$ emission is too weak to be identified, and instead, the Yb³⁺ emission is dominant. This is due to a highly efficient energy transfer from Er³⁺ ${}^{4}I_{11/2}$ to Yb³⁺ ${}^{2}F_{5/2}$, 14 which is also evidenced by the very weak 1530 nm emission of Er³⁺ under 650 nm excitation. In addition, one may find that the Yb³⁺ emission intensities decrease with increasing *x* in the *x* range of 0.1–0.4 for 520 nm excitation. The decline is due to concentration quenching because the critical concentration of Yb³⁺ was observed to be 7.36% in the Y₂O₃ host.¹⁵

We evaluate η_{r4} , η_{MPR43} , η_{CRNB} and η_{CRB} based on the observed increase in the Er³⁺ 1530 nm emission, Yb³⁺ 1040 nm emission and Er³⁺ R/G ratio for 520 nm excitation. Considering the distinct increment $(I_1 - i_1)$ in 1530 nm emission intensity in the presence of Yb³⁺, one may also observe a small increase for x=0. The small increase is attributed to the feeding by the ${}^{4}S_{3/2} \rightarrow {}^{4}I_{13/2}$ radiative transition, which has a big branch ratio $(\beta=0.27)$.¹⁶ Hence both the CRNB and the ${}^{4}S_{3/2} \rightarrow {}^{4}I_{13/2}$ radiative transition contribute to the increase. Regarding the experimentally observed *x*-independent ${}^{4}I_{13/2}$ lifetime, we have, as described in Supplementary Equations (S1)–(S4)

$$\frac{\eta_{\rm CRNB}}{\beta \eta_{r_4}} = \frac{I_1 - i_1}{I_4} \left(\frac{I_4}{I_1 - i_1} \right)_0 - 1 \tag{1}$$

where I_i and i_i are emission intensities from the *i*th state of Er^{3+} upon 520 nm and 650 nm excitations, respectively. These emission intensities on the right side of Equation (1) are dependent on *x*.

For the increment $(I_d - i_d)$ in Yb³⁺ emission intensity, the CRNB is the only contributor, while the rest intensity (i_d) is populated by quasiresonant energy transfer from Er^{3+ 4}I_{11/2} with a transfer efficiency close to 1 due to the high Yb³⁺ concentration. The ⁴I_{11/2} can be populated from the ⁴F_{9/2} via the ⁴I_{9/2} by cascade MPR with an efficiency of almost 100%, regarding a very short ⁴F_{9/2} lifetime (approximately 20 µs) compared with its radiative lifetime (600 µs).¹⁶ As a result, the population of the ⁴F_{9/2} by the MPR and the CRB from the ⁴S_{3/2} finally and completely reaches the ⁴I_{11/2}. Hence, the excitation source of the rest Yb³⁺ emission is the MPR and the CRB from the ⁴S_{3/2}. Then, we have

$$\frac{\eta_{\rm CRNB}}{\eta_{\rm MPR_{43}} + \eta_{\rm CRB}} = \frac{I_{\rm d} - i_{\rm d}}{i_{\rm d}} \tag{2}$$

where the subscript d denotes Yb^{3+} .

The red emission is populated by MPR and CRB from the ${}^{4}S_{3/2}$, so the R/G ratios simply satisfy

$$\frac{R}{G} = \left(\frac{R}{G}\right)_0 \left(1 + \frac{\eta_{CRB}}{\eta_{MPR_{43}}}\right) \frac{\tau_3}{\tau_{3,0}}$$
(3)

Using Equations (1)–(3) with $\eta_{r4}+\eta_{MPR43}+\eta_{CRNB}+\eta_{CRB}=1$, we calculated the efficiencies, as shown in Figure 4a. We obtained $\eta_{MPR43}/\eta_{r4}=2.08$, which yields $\eta_{r4}=32\%$ for cubic Y₂O₃:Er³⁺. The measured green PL intensities (square) follow the tracks of η_{r4} well, but it decreases faster than τ_4 with increasing *x* (Table 1). This feature further implies the existence of an undetected fast decay in the green emission, as indicated in the above section, like the case of the Perrin model.¹⁷ We may consider an 'active sphere' centered at an Er³⁺ ion so that its green emission can be completely quenched by a Yb³⁺ ion located within the sphere through the CR. If the number of the effective cation sites within the sphere is *n*, the fraction (*f*) of Er³⁺ ions that



Figure 4 (a) Area graph for efficiencies of various processes for depopulating the ${}^{4}S_{3/2}$ state in Y₂O₃:0.002Er³⁺, xYb³⁺ under 520 nm excitation. The measured green emission intensities are also shown (squares). (b) Comparison of experimental and theoretical dependence of *f* on *x*. The experimental data are obtained by $(I_{4}/I_{4,0})(\tau_{4,0}/\tau_{4})$ (circle) or $(\eta_{r4}/\eta_{r4,0})(\tau_{4,0}/\tau_{4})$ (filled circle).

have no Yb³⁺ in the sphere satisfies the binomial distribution: $f=(1-x/2)^n$. Considering that the green luminescence yield (I_4 or η_{r4}) can be regarded as the area under its decay curve including the undetected fast component, then f can be obtained either by ($I_4/I_{4,0})(\tau_{4,0}/\tau_4)$ or ($\eta_{r4}/\eta_{r4,0})(\tau_{4,0}/\tau_4)$ from the decay picture of the green emission containing a fast component with a lifetime close to 0 and a slow one with a lifetime of τ_4 , weighted by 1-f and f, respectively, as described in Supplementary Fig. S1 and Equation (S5).

In cubic Y_2O_3 , there are two Y sites with respective C_{3i} and C_2 symmetries. The C_{3i} has an inversion center on this site; the 4f-4ftransition of a rare earth ion is forbidden. Therefore, only the C_2 site is effective. A C_2 site has four of the first nearest C_2 sites, four of the second nearest C_2 sites and six of the next nearest C_2 sites. The fitting results point to n=8, as shown in Figure 4b, indicating that the green emission of an Er³⁺ can be completely quenched only by an Yb³⁺ located in the nearest site of the Er³⁺ ion. As a result, the nearest Er³⁺-Yb³⁺ pairs exhibit completely quenched green emission and produce the undetected fast decay of the green emission. As we have observed, the CRB is a fast process; thus, the CRB takes place only within the nearest $Er^{3+} - Yb^{3+}$ pairs. Of course, the occurrence of a fast CRNB is also expected in the nearest Er³⁺-Yb³⁺ pairs. Accordingly, the distant Er³⁺-Yb³⁺ pairs can only perform the slow CRNB to reduce τ_4 . In Figure 4a, over 30% of the CR is the CRB upon 520 nm excitation, and the CRB becomes more important than the MPR for populating the red-emitting state from the green state as $x \ge 0.1$. The efficient CRB benefits from the nearest $\text{Er}^{3+} - \text{Yb}^{3+}$ pairs that can compete with the energy transfer from the Yb³⁺ ions excited by the CR to other Yb^{3+} ions outside the pairs. The nearest $Er^{3+}-Yb^{3+}$ pair ensures that the CRB takes place within the pair because the

energy back transfer from Yb³⁺ to other Er³⁺ undergoes a longer interaction distance due to the low Er^{3+} concentration used in this work. Hence, no dependency of the CRB on excitation densities is observed in the present work. The Perrin model, like CRB, may be governed by the exchange mechanism described by Inokuti and Hirayama.¹⁷

In CRB, the CR transfer from Er^{3+} (${}^{4}\mathrm{S}_{3/2} \rightarrow {}^{4}\mathrm{I}_{13/2}$) to Yb^{3+} (${}^{2}\mathrm{F}_{5/2} \leftarrow {}^{2}\mathrm{F}_{7/2}$) and the back transfer from Yb^{3+} (${}^{2}\mathrm{F}_{5/2} \rightarrow {}^{2}\mathrm{F}_{7/2}$) to Er^{3+} (${}^{4}\mathrm{F}_{9/2} \leftarrow {}^{4}\mathrm{I}_{13/2}$) suffer from an energy mismatch of ~1700 cm⁻¹ and ~1500 cm⁻¹, respectively, which is more than twice as large as the cutoff phonon energy of 600 cm⁻¹ in Y₂O₃. The performance of the energy transfers thus requires emission of two or three phonons to make up for the large energy mismatch. In spite of requirement of several phonons, the high Yb³⁺ concentration and the nearest Er^{3+} – Yb³⁺ pairs make the transfers efficient.

Effect of the CRB on UCL

Figure 5 shows the UCL spectra for $Y_2O_3:0.002Er^{3+}$, $0.1Yb^{3+}$ upon pulse and CW infrared (980 nm) excitations. For comparison, its PL spectrum is also presented. The phenomenon^{11,18–20} that the R/G ratio in PL is much less than that in UCL is evident. Ignoring the CRB, it was naturally deduced from the phenomenon that the ETU from ${}^{4}I_{13/2}$ makes the major contribution to the red UCL.^{18,19} If the CRB is operative, the above deduction could be possibly invalid because of the following reasons: the nearest Er³⁺-Yb³⁺ pairs can be preferentially excited to the Er^{3+ 4}S_{3/2} state by two sequential energy transfers from Yb³⁺ in ETU, as shown in the insert to Figure 5. Meanwhile, the CRB enables the nearest $Er^{3+}-Yb^{3+}$ pairs to only have a red emission in the visible spectral region. In the PL measurement, however, each Er³⁺ ion has the same probability to be excited to the ⁴S_{3/2} state by ground state absorption. Based on the analysis above, a much smaller R/G ratio of PL than that of UCL is still achievable if the CRB makes the major contribution to the red UCL. In other words, the observation of a much smaller R/G ratio for PL than for UCL cannot exclude the major role of the ⁴S_{3/2} state in populating the red-emitting state in the upconversion of the Er3+-Yb3+ system. In our experiment, the PL and UCL temporal behaviors of Y₂O₃:0.002Er³⁺, 0.1Yb³⁺ (Figure 6)



Figure 5 UCL spectra upon CW and upon pulse 980 nm excitations as well as a PL spectrum for $Y_2O_3:0.002Er^{3+}$, $0.1Yb^{3+}$. The green emission intensities are normalized. The CW excitation density is as low as 50 mW cm⁻². The insert is the diagram of ETU processes for the $Er^{3+}-Yb^{3+}$ system. ETU, energy transfer upconversion; PL, photoluminescence; UCL, upconversion luminescence.



Figure 6 Time evolutions of the green UCL (560 nm), the red UCL (670 nm), Yb^{3+} emission (1040 nm) and Er^{3+} emission (1530 nm) from ${}^{4}l_{13/2}$ after pulse 980 nm excitation for Y_2O_3 :0.002 Er^{3+} , 0.1 Yb^{3+} . UCL, upconversion luminescence.

demonstrate that the ${}^{4}F_{9/2}$ is populated mainly from ${}^{4}S_{3/2}$ in ETU, as analyzed below.

From Figure 6, the Yb³⁺ emission decays exponentially (with a decay time of τ_d =840 µs), indicating the rapid diffusion limited energy transfer. The decay time of the green UCL is 391 µs, and that of the red is 412 μ s. These UCL decay times are much longer than τ_4 (66 μ s) and τ_3 (20.9 μ s) (Table 1), exhibiting the typical effect of ETU. Thus, the UCL decay function is mainly determined by the product of the decay functions of the Yb^{3+ 2} $F_{5/2}$ and Er^{3+} intermediate states. The green UCL decay time approaches $\tau_d/2$, reflecting that $\mathrm{Er}^{3+4}I_{11/2}$, as the intermediate state for the green upconversion, has the same decay time as $Yb^{3+2}F_{5/2}$ because the two levels are thermally coupled in case of rapid energy transfer between them, as predicated earlier.¹⁴ The approximation of the red UCL decay time to the green one indicates that the ${}^{4}F_{9/2}$ is populated mainly from the ${}^{4}S_{3/2}$ under pulse infrared excitation; otherwise, the red could have a decay time close to τ_d (840 μ s) because of the quite long lifetime (τ_1 =8.4 ms) of the intermediate ${}^{4}I_{13/2}$ state. Now, we want to determine if the ${}^{4}F_{9/2}$ is still populated mainly from the ⁴S_{3/2} under CW infrared excitation.

In Figure 5, one may find that the R/G ratio in UCL for CW excitation, $(R/G)_{UCL}@CW$, is larger than that for pulse excitation, $(R/G)_{UCL}@pulse$. The smaller $(R/G)_{UCL}@pulse$ arises from the low build-up process of the ${}^{4}I_{13/2}$ state (Figure 6), as described in Supplementary Equation (S10), which slows down the ETU from the ${}^{4}I_{13/2}$ to the ${}^{4}F_{9/2}$ after pulse infrared excitation. Taking the different R/G ratios into account, the percentages of the red component R_4 , populated from the ${}^{4}S_{3/2}$, and R_1 , from the ${}^{4}I_{13/2}$, can be evaluated.

If the ${}^{4}S_{3/2}$ and the ${}^{4}F_{9/2}$ states are populated under infrared excitation in ETU, we denote the mean emitting efficiencies of the two states by η_{G} and η_{R} , respectively. While the mean depopulating efficiency from the ${}^{4}S_{3/2}$ to the ${}^{4}F_{9/2}$ via MPR and the CRB is denoted by η_{43} , the R/G ratio in UCL is

$$\left(\frac{\mathbf{R}}{\mathbf{G}}\right)_{\mathrm{UCL}} = \frac{(1+R_1/R_4)\eta_{43}\eta_{\mathbf{R}}}{\eta_{\mathbf{G}}} \tag{4}$$

where η_G , η_R and η_{43} are unchanged for CW or pulse infrared excitations. In view of the high Yb³⁺ and low Er³⁺ concentrations, we assume here the simplest possible model; (i) ETU undergoes the rapid diffusion limited energy transfer from Yb³⁺ that can be described by an average energy transfer rate; (ii) the ground and excited states absorption of Er^{3+} is neglected. We also assume the ${}^{4}\mathrm{F}_{7/2}$ can rapidly and completely relax down to the ${}^{4}\mathrm{S}_{3/2}$ state due to their proximity in energy.

From the diagram of ETU processes inserted in Figure 5, the population flow from the ${}^{4}I_{11/2}$ to the ${}^{4}F_{7/2}$ is $C_{d5}n_dn_2$ and that from the ${}^{4}I_{13/2}$ to the ${}^{4}F_{9/2}$ is $C_{d3}n_dn_1$. Under CW excitation, the population flow is time independent, and thus, $R_4@CW = \eta_R\eta_{43}C_{d5}n_dn_2$ and $R_1@CW = \eta_R C_{d3}n_dn_1$, where C_{di} is the coefficient for energy transfer from donor Yb³⁺ to the *i*th state of Er³⁺, and n_d and n_i are populations of Yb³⁺ ${}^{2}F_{5/2}$ and Er³⁺ in the *i*th state, respectively. Then, the R_1/R_4 ratio for CW infrared excitation is expressed as

$$(R_1/R_4)@CW = (C_{d3}/C_{d5})W_{21}\tau_1/\eta_{43}$$
(5)

where W_{21} is the sum of the ${}^{4}I_{11/2} - {}^{4}I_{13/2}$ radiative and MPR rates, which satisfies $W_{21}n_2 = n_1/\tau_1$ for a steady state excitation.

Under pulse infrared excitation of Yb^{3+} , we detect the upconversion luminescence yield over time. Based on the rate equations for describing ETU in the $Er^{3+}-Yb^{3+}$ system, as in Supplementary Equations (S7)–(S13), we obtain

$$(R_1/R_4)@pulse = [\tau_d/(\tau_1 + \tau_d)](C_{d3}/C_{d5})W_{21}\tau_1/\eta_{43}$$
(6)

From Figure 5, the R/G ratio in UCL for CW excitation is 1.2 times as large as that for pulse excitation,

$$(\mathbf{R}/\mathbf{G})_{\mathrm{UCL}}@\mathrm{CW} = 1.2(\mathbf{R}/\mathbf{G})_{\mathrm{UCL}}@\mathrm{pulse}$$
(7)

Combining Equations (4)–(7) with τ_d =840 µs and τ_1 =8.4 ms gives (R_1/R_4) @pulse=0.02 and (R_1/R_4) @CW=0.22 for Y₂O₃:0.002Er³⁺, 0.1 Yb³⁺. These values mean that only 2% of the red UCL is populated from the ⁴I_{13/2} and 98% from the ⁴S_{3/2} under pulse infrared excitation, and for CW infrared excitation, 18% of the red UCL is populated from the ${}^{4}I_{13/2}$ and 82% from the ${}^{4}S_{3/2}$. The main role of the ${}^{4}S_{3/2}$ in populating the red UCL reflects a strong CRB. Meanwhile, the small contribution of the ETU from the ⁴I_{13/2} is partially due to a small value of C_{d3}/C_{d5} that is calculated to be $0.124\eta_{43}$ using Equation (5). In this calculation, a W_{21} of 218 s⁻¹ is used, which is determined from our measured ${}^{4}I_{11/2}$ lifetime ($\tau_{2,0}$) of 2.73 ms in Y₂O₃:0.002Er³⁺ and the reported¹⁶ intrinsic ${}^{4}I_{11/2}$ lifetime of 6.03 ms and ${}^{4}I_{11/2} \rightarrow {}^{4}I_{13/2}$ radiative branch ratio of 0.108. The small C_{d3}/C_{d5} ratio is well understood in view of a large energy mismatch (${\sim}1500\,\,{\rm cm}^{-1})$ in the transfer from $Yb^{3+}({}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2})$ to $Er^{3+}({}^{4}F_{9/2} \leftarrow {}^{4}I_{13/2})$ relating to C_{d3} , while the energy transfer from $Yb^{3+}({}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2})$ to $Er^{3+}({}^{4}F_{7/2} \leftarrow {}^{4}I_{11/2})$ relating to C_{d5} is quasi-resonant.

Obviously, a high Yb³⁺ concentration and preferential excitation of the nearest $Er^{3+}-Yb^{3+}$ pairs in ETU can promote the CRB process. Meanwhile, increasing the population ratio of $Er^{3+} \, {}^4I_{11/2}$ to ${}^4I_{13/2}$ is beneficial to ETU from the ${}^4I_{11/2}$ to the ${}^4F_{7/2}$, and it thus can promote the contribution of the CRB to the red UCL against the ETU from the ${}^4I_{13/2}$ to the ${}^4F_{9/2}$. Therefore, the host material with low cutoff phonon energy likely has a pronounced CRB process like Y_2O_3 . However, this population ratio is usually reduced in nanomaterials because the ${}^4I_{11/2} - {}^4I_{13/2}$ MPR is strongly enhanced by coupling with OH⁻ groups on the large surface of the nanomaterials.²¹

CONCLUSIONS

We observed a pronounced CRB process for populating the Er^{3+} redemitting state from its green state in cubic Y_2O_3 : Er^{3+} , Yb^{3+} . We found the CRB can be more efficient than MPR and can even make the major contribution to the red emission in both PL and UCL. The CRB takes place only in the nearest Er^{3+} – Yb^{3+} pairs, and thus, it is a fast and 6

efficient process at low excitation densities. The present research methods may be applied to a general $\mathrm{Er}^{3+}-\mathrm{Yb}^{3+}$ system, and the identification of the importance of the CRB in other material hosts is expected.

ACKNOWLEDGEMENTS

This work has been supported by NSFC (Grants No. 10834006, 51172226, 61275055 and 11274007).

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