

Journal of Alloys and Compounds 309 (2000) 10-15



www.elsevier.com/locate/jallcom

High-resolution photoluminescence spectrum of GdVO₄:Eu^{3+☆}

Qing-li Zhang a,b, Chang-xin Guo b, Chao-shu Shi b, Shao-zhe Lüc

^aStructure Research Lab., University of Science and Technology of China, 230026, Hefei, China ^bDepartment of Physics, University of Science and Technology of China, 230026, Hefei, China ^cOpen Laboratory of Changchun Institute of Physics, Academia Sinica, 130021, Changchun, China

Received 4 February 2000; accepted 26 May 2000

Abstract

In this article, the high-resolution photoluminescence spectrum of $GdVO_4$: Eu^{3+} was studied. The result indicates that Eu^{3+} in $GdVO_4$: Eu^{3+} replaces Gd^{3+} and occupies D_{2d} symmetry, the position 4a (000) of the space group I4/amd. The decay times of the $^5D_0 \rightarrow ^7F_2$ and $^5D_1 \rightarrow ^7F_1$ emissions of Eu^{3+} in $GdVO_4$: Eu^{3+} were measured, they are 0.44 ms and 10.7 μ s, respectively. The infrared spectrum of $GdVO_4$: Eu^{3+} consists of two peaks at 843.54 and 451.45 cm⁻¹ and indicates that an Eu^{3+} in the state 5D_1 can relax to the state 5D_0 by emitting two phonons. © 2000 Elsevier Science S.A. All rights reserved.

Keywords: GdVO₄:Eu³⁺; Photoluminescence; High-resolution spectrum; Decay time; Infrared spectrum

PACS: 78.55

1. Introduction

GdVO₄ is an excellent laser medium. It has attracted great interest recently. The studies of GdVO4 doped with rare earth ions Pr³⁺, Nd³⁺, Eu³⁺, Ho³⁺, Er³⁺, Tm³⁺ and Yb³⁺ have been reported [1–7]. GdVO₄:Eu³⁺ is an interesting red-emitting material [5-7]; it's most important application would be as a laser material. Because of a strong absorption to ultraviolet light by GdVO4, an effective energy transfer from VO_4^{3-} to Eu^{3+} and effective excitation of Eu³⁺ by the 450-nm strong emission of VO₄³⁻, GdVO₄:Eu³⁺ is a highly efficient red-emitting material. Under 253.7-nm excitation, YVO₄:Eu³⁺, LuVO4:Eu³⁺, Y₂O₂S:Eu³⁺ and GdVO₄:Eu³⁺ has almost equal emission intensity. At the same time, we found that GdVO₄:Eu³⁺ has a very good temperature property, from a few K to room temperature its principal emission has no obvious change, above the room temperature, its principal emission increases with increasing temperature and by more than one order at about 600 K. Thus, it can be used in high-pressure mercury light etc. The principal emission peak of GdVO₄:Eu³⁺ is 619 nm, according to our calcula-

are studied.

The polycrystalline sample of $GdVO_4$: Eu^{3^+} was prepared by reacting stoichiometric proportions of V_2O_5 and Gd_2O_3 with purity 99.99 and 99.95%, respectively, and the nitric acid solution of Eu_2O_3 was used to dope Eu^{3^+} into $GdVO_4$. The stoichiometric mixture was sintered in air atmosphere at $800^{\circ}C$ for 8 h and the polycrystalline $GdVO_4$: Eu^{3^+} was obtained.

tion, its colourity coordinates are x=0.67 and y=0.32. It

been reported [5,6], but its high-resolution photoluminesc-

ence of 0.15 cm⁻¹ and by which the symmetry of Eu³⁺ replacing Gd³⁺ in the lattice is shown. Infrared spectrum

and decay time of Eu³⁺ emission have not been reported before to our knowledge. In this article, these properties

GdVO₄ has ZrSiO₄ structure, belonging to the space

group I4/amd (Gd³⁺, V⁵⁺ and O²⁻ occupy the positions $4a(0\ 0\ 0)$, $4b\ (0\ 0\ 0.5)$ and $16\ h\ (0\ x\ z)$, respectively) [8]. It

is tetragonal system and its lattice parameters are a = 7.2176 Å, c = 6.3483 Å, Z = 4, $D_x = 5.474 \text{ g/cm}^3$ [9].

Some luminescence properties of GdVO₄:Eu³⁺ have

can be used as trichromatic material.

In order to determine the crystal structure of GdVO₄,

E-mail address: shics@utsc.edu.cn (C.-s. Shi).

0925-8388/00/\$ – see front matter © 2000 Elsevier Science S.A. All rights reserved. PII: S0925-8388(00)01051-3

^{2.} Experiments

^{*}Supported by NSFC of China under Grant No. 59732040.

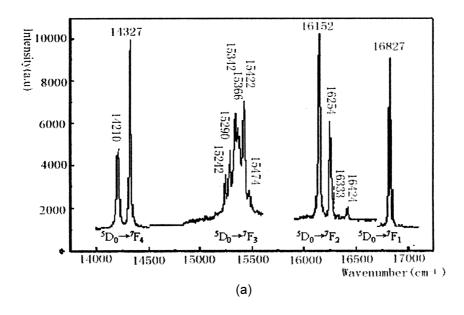
^{*}Corresponding author.

its X-ray diffraction measurement was performed at room temperature using a D/Mrax-rA Rotating Anode X-ray Diffractometer with CuK α radiation. The position and intensity of diffraction peaks of the polycrystalline powder GdVO₄ are consistent with that of the powder diffraction file (PDF) 17-260, which indicates that the sample crystalized well. From the experimental data, the lattice parameters a=7.22Å and c=6.38 Å were calculated with the least square method. From the formula D_x =1.66 ZM_r/V (where V and Z are the volume and number of molecules of a primitive unit cell, M_r is the molecular weight, respectively), the density of GdVO₄ D_x =5.46 (g/cm³) was calculated and consistent with the D_x data of PDF 17-260.

The high-resolution photoluminescence of GdVO₄:Eu³⁺ was excited by the 266-nm laser line of the fourth order output of a GCR-2A YAG:Nd laser. The monochromator, with resolution 0.15 cm⁻¹, is a SPEX-1403 double raster monochromator of SPEX corporation. The high-resolution photoluminescence spectrum is shown in Fig. 1.

The luminescence decay times of Eu³⁺ in GdVO₄:Eu³⁺ were measured at room temperature using a 162 BOX-CAR.

The infrared spectrum of GdVO₄:Eu³⁺ was performed on a Fourie Transform Infrared Spectrometer of Nicolet Instrument. Its maximal resolution is 0.1 cm⁻¹ and accuracy is superior to 0.01 cm⁻¹. Before measuring, the mixture of GdVO₄:Eu³⁺ and KBr with fluorescence purity was



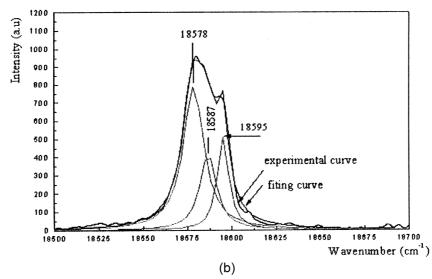


Fig. 1. High-resolution photoluminescence spectrum from (a) $^5D_0 \rightarrow ^7F_J$ (J=1, 2, 3, 4) transition and (b) $^5D_1 \rightarrow ^7F_1$ transition of GdVO₄:Eu³⁺(10 mol%) ($\lambda_{\rm ex} = 266$ nm).

Table 1 Lineshape fitting parameters of the spectrum curve of ${}^5D_1 \rightarrow {}^7F_1$ transition of Eu³⁺ in GdVO₄

Spectrum line center (cm ⁻¹)	Width (cm ⁻¹)	Height (a.u)
18 578	14.3	787.5
18 587	11.3	423.4
18 595	7.8	514.1

ground in an agate, then, the mixture was pressed into a disk with a tablet machine and was measured.

3. Experimental results and discussion

3.1. High-resolution photoluminescence spectrum of $GdVO_4$:Eu³⁺

The high-resolution photoluminescence spectrum (HRPLS) of $GdVO_4$: Eu^{3+} is shown in Fig. 1. The spectrum shape of ${}^5D_1 {\rightarrow} {}^7F_1$ transition of Eu^{3+} was fitted with Lorentz lineshape, as shown in Fig. 1b and the fitting parameters are listed in Table 1. If the spectrum curve is fitted with Guass lineshape, there will be a very wide and flat peak, which does not seem reasonable. The lineshape of the ${}^5D_1 {\rightarrow} {}^7F_1$ transition is a Lorentz lineshape, which maybe indicate that the linear broadening in $GdVO_4$: Eu^{3+} was mainly caused by thermal vibration of the lattice.

Because Gd³⁺ and Eu³⁺ are neighbor elements on the

periodic table, their ion radii are 0.95 and 0.94 Å, respectively. It is commonly suggested that Eu^{3+} replaces Gd^{3+} in GdVO_4 : Eu^{3+} . After replacing Gd^{3+} , Eu^{3+} will be surrounded by eight O^{2-} and has $\mathrm{D}_{2\mathrm{d}}$ symmetry, which results in energy level splitting of Eu^{3+} and the photoluminescence spectrum of Eu^{3+} will be obviously different from that of Eu^{3+} in the crystal fields belonging to the other symmetry. Calculated by crystal field theory [10,11], the energy level splitting of Eu^{3+} in symmetry $\mathrm{D}_{2\mathrm{d}}$ was shown in Tables 2 and 3. In Tables 2 and 3, the theoretical and experimental ${}^5\mathrm{D}_0 \!\to^7\!\mathrm{F}_{\mathrm{J}}$ ($J\!=\!1,\ 2,\ 3,\ 4$) and ${}^5\mathrm{D}_1 \!\to^7\!\mathrm{F}_1$ transition values also have been listed.

From Tables 1 and 2 it can be seen that the observed lines are consistent with the calculated lines very well, except that the lines 14 275, 14 375 cm⁻¹ of $^5D_0 \rightarrow ^7F_4$ transition and 16 843 cm⁻¹ of $^5D_0 \rightarrow ^7F_1$ transition are too weak to emerge.

The lines 16 424 and 16 333 cm $^{-1}$ near the $^5D_0 \rightarrow ^7F_2$ transition of Eu $^{3+}$ are from $^5D_2 \rightarrow ^7F_6$ transition of Eu $^{3+}$. The lines 15 366, 15 422 and 15 474 cm $^{-1}$ near the $^5D_0 \rightarrow ^7F_3$ transition of Eu $^{3+}$ are from the $^4G_{5/2} \rightarrow ^6H_{9/2}$ transition of Sm $^{3+}$, which exists as the impurity in Eu $_2O_3$. Because of the high concentration of Eu $^{3+}$, it is possible that a small amount of Sm $^{3+}$ was taken into the sample from Eu $_2O_3$. On the other hand, $^5D_0 \rightarrow ^7F_3$ emission of Eu $^{3+}$ is very weak, so even if the principal emission of Sm $^{3+}$ is very weak, they have almost equal intensity and can be detected at the same time.

As shown in Fig. 2, the photoluminescence spectrum

Table 2 Energy level 7F_1 and $^5D_0 \rightarrow ^7F_1$ transition of Eu $^{3+}$ in D_{24} symmetry in GdVO $_4$

J	Γ	Energy level (cm ⁻¹)	$^{5}D_{0} \rightarrow ^{7}F_{J}$ transition			
			ED	MD	Experimental values (cm ⁻¹)	Calculated values (cm ⁻¹)
0	A_1		_	_		
1	\mathbf{A}_2	350	_	+	Too weak to emerge	16 843
	E	368	+	+	16 827	16 825
2	\mathbf{B}_{1}	1122	_	_		
	E	1044	+	+	16 152	16 149
	\mathbf{A}_1	993	-	-		
	\mathbf{B}_2	935	+	_	16 254	16 258
3	E	1954	+	+	15 242	15 239
	$\mathbf{B}_{_{1}}$	1910	-	-		
	\mathbf{A}_2	1906	-	+	15 290	15 287
	E	1864	+	+	15 342	15 329
	\mathbf{B}_2	1854	+	_	15 342	15 339
4	$\mathbf{A}_{_{1}}$	3062	_	_		
	E	2987	+	+	14 210	14 206
	\mathbf{A}_2	2918	_	+	Too weak to emerge	14 275
	A_1	2884	-	-		
	\mathbf{B}_2	2866	+	-	14 327	14 326
	E	2818	+	+	Too weak to emerge	14 375
	$\mathbf{B}_{_{1}}$	2701				

^a G, ED and MD stand for unreducible representation, electronic and magnetic dipole transition, respectively. + and - stand for transition permitted and forbidden. The energy level 5D_0 is 17 193 cm⁻¹.

^b Refs. [10,11].

Table 3 $^5D_1 \rightarrow ^7F_1$ transition of Eu $^{3+}$ in D_{2d} symmetry in GdVO $_4$ ab

Initial state energy level (cm ⁻¹)		Final state energy level (cm ⁻¹)		5 D ₁ \rightarrow 5 F ₁ transition	
⁵ D ₁		7D ₁		Experimental values (cm ⁻¹)	Calculated values (cm ⁻¹)
$\overline{A_2}$	18 947	A_2	368	_	
		E	350	18 587	18 597
E	18 949	A_2	368	18 578	18 581
		E	350	18 595	18 599

^a Where '-' stands for the transition forbidden.

(PLS) of $GdVO_4$: Eu^{3+} and $GdVO_4$: Sm^{3+} was measured with a 850 fluorescence spectrophotometer of HITACHI corporation.

From the photoluminescence spectrum of GdVO₄:Sm³⁺ (Fig. 2b), it can be seen that the strongest emission of GdVO₄:Sm³⁺ peaks at about 603.7, 606.8 and 646.4, 654.6 nm, which are from ${}^{4}G_{5/2} \rightarrow {}^{6}H_{7/2}$ and ${}^{4}G_{5/2} \rightarrow {}^{6}H_{9/2}$ transitions of Sm³⁺, respectively. The Sm³⁺ lines emerging in HRPLS of GdVO₄:Eu³⁺ (Fig. 1a) are the lines 650.8, 648.4 and 646.3 nm (15 366, 15 422 and 15 474 cm⁻¹, respectively), corresponding to the lines 654.6 and 646.4 nm of PLS of GdVO₄:Sm³⁺, respectively. We can see that the lines 650.8 and 648.4 nm of HRPLS were not discerned by the 850 fluorescence spectrophotometer. They correspond to the same line at 654.6 nm in PLS of GdVO₄:Sm³⁺. Because the luminescence efficiency of GdVO₄:Sm³⁺ is not so high that the slits of the fluorescence spectrophotometer can not be opened very narrowly, which results in the two lines not been discerned by the 850 fluorescence spectrophotometer. The lines of GdVO₄:Sm³⁺ in HRPLS and in PLS are very consistent, so we can conclude that the lines 15 366, 15 422 and $15 474 \text{ cm}^{-1}$ are from the emission of the impurity Sm^{3+} .

If the symmetry of Eu³⁺ is D_{2d} , its 580-nm emission from ${}^5D_0 \rightarrow {}^7F_0$ transition is forbidden as for magnetic or electronic dipole transition. It is consistent with the results shown in Fig. 2, the 580 nm (that is 17 241 cm⁻¹) transition didn't emerge.

From the above, it can be concluded that Eu^{3+} replaces Gd^{3+} , has D_{2d} symmetry and occupies the position $4a(0\ 0\ 0)$ of the space group I4/amd.

3.2. Luminescence decay time of GdVO₄:Eu³⁺

The luminescence decay curves of $GdVO_4$: Eu^{3+} are shown in Fig. 3 (solid curves), which were fitted by $I=I_0\exp{(-t/\tau)}$ (where I_0 and t denote initial luminescence intensity and the time when luminescence intensity decays to I_0/e , the fitting curves are dot curves). The fitting parameters are listed in Table 4. The decay times of ${}^5D_0 \rightarrow {}^7F_2$ and ${}^5D_1 \rightarrow {}^7F_1$ emissions are 0.44 ms and 10.72 μ s, respectively. From the exponent decay law of the

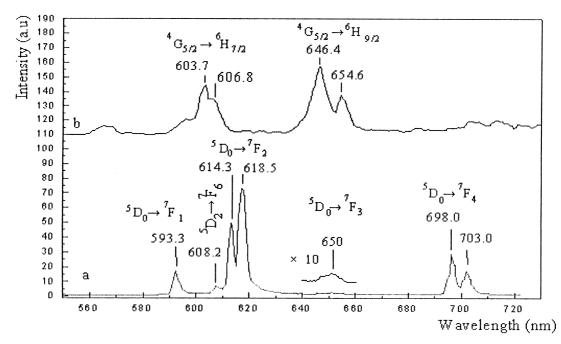


Fig. 2. Photoluminescence spectrum of (a) $GdVO_4$: $Eu^{3+}(10 \text{ mol }\%)$ and (b) $GdVO_4$: $Sm^{3+}(1 \text{ mol }\%)$ ($\lambda_{ex} = 310 \text{ nm}$).

^b Refs. [11,12].

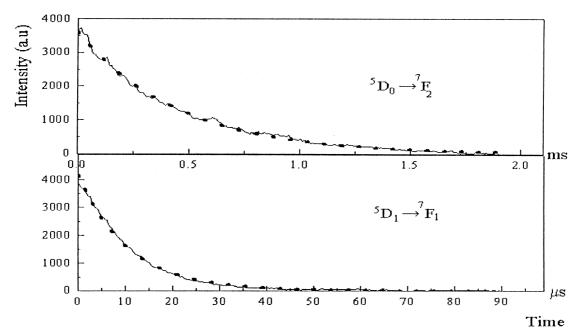


Fig. 3. Luminescence decay curves of ${}^5D_1 \rightarrow {}^7F_1$ and ${}^5D_0 \rightarrow {}^7F_2$ emissions of GdVO₄:Eu³⁺(10 mol %) (λ_{ex} = 266 nm). In the figure, solid and dot curves are experimental and fitting curves, respectively.

emission, Eu³⁺ in GdVO₄:Eu³⁺ is an isolated luminescence center.

But the decay time of ${}^5D_1 \rightarrow {}^7F_1$ emission is more than 40 times faster than that of the ${}^5D_0 \rightarrow {}^7F_2$ emission, due to the interaction between Eu³⁺ and lattice. A rare earth ion in an excited state i decays with a mean lifetime τ_i given by

$$\frac{1}{\tau_{\rm i}} = \sum_{\rm j} (A_{\rm ij} + W_{\rm ij})$$

where A and W are the probabilities for radiative and nonradiative decay, respectively, and the summation is over all terminal states j. A and W are the functions of energy difference [13,14]. For J manifolds separated by small energy gaps, ΣW_{ij} is usually much greater than ΣA_{ij} . With respect to the energy level 5D_1 of Eu $^{3+}$, there is an energy level 5D_0 , which is only 1740 cm $^{-1}$ below it, so through the interaction with the lattice, the excited state 5D_1 can relax to the state 5D_0 through nonradiative transition quickly, which results in the mean lifetime of 5D_0 being very short. As for the excited state 5D_0 , the nearest energy level below it is 7F_6 , the energy difference

Table 4 Fitting parameters of luminescence decay curves of $^5D_1 \rightarrow ^7F_1$ and $^5D_0 \rightarrow ^7F_2$ emissions^a

	$^{5}D_{1} \rightarrow ^{7}F_{1}$	$^{5}D_{0} \rightarrow ^{7}F_{2}$
I_0 (a.u)	4158.6	3628.4
τ (µs)	10.72	0.44

 $^{^{\}rm a}I_0,\, au$ are initial luminescent intensity and the time when luminescence intensity decays to $I=I_0/e$, respectively.

between 5D_0 and 7F_6 state is 16 155 cm $^{-1}$, which is far greater than the energy of a phonon, this results in it being hardly possible that the multiphonon transition from ${}^5D_0 \rightarrow {}^7F_J$ occurs, so nonradiative transition probability $\Sigma W_{ij} << \Sigma A_{ij}$, the mean lifetime of the excited state 5D_0 becomes long.

3.3. Infrared transmission spectrum of GdVO₄:Eu³⁺

The infrared spectrum of GdVO₄:Eu³⁺ is shown in Fig. 4. There are two peaks at 834.54 and 451.45 cm⁻¹. Through a Fourie self-deconvolution method program, which is provided by the software of Nicolet Instrument, the peak at 834.54 cm⁻¹ was decomposed into seven peaks, which are at 949.89, 881.57, 856.15, 834.90, 808.59, 785.66 and 770.44 cm⁻¹, respectively, as shown in the small box of the Fig. 4. These peaks show the absorption of the lattice phonon. In GdVO₄:Eu³⁺, the energy difference between ⁵D₀ and ⁵D₁ is 1740 cm⁻¹, which is double 870 cm⁻¹ and in the range of 949.89 – 770.44 cm⁻¹. So in GdVO₄:Eu, an electron in the state ⁵D₁ can relax to the state ⁵D₀ quickly by emitting two phonons of about 870 cm⁻¹.

4. Conclusion

1. The high-resolution photoluminescence spectrum of GdVO₄:Eu³⁺ indicates that Eu³⁺ replaces Gd³⁺, has D_{2d} symmetry and occupies the position 4a(0 0 0) of the space group *I4/amd*.

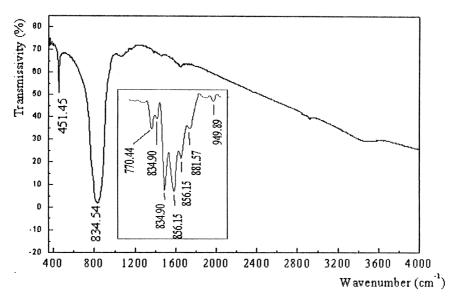


Fig. 4. Infrared spectrum of $GdVO_4$: $Eu^{3+}(10 \text{ mol}\%)$. The curve in the small box of the figure is the peaks obtained by decomposing the peak 834.54 cm⁻¹ with the Fourie self-deconvolution method.

- 2. The decay times of ${}^5D_0 \rightarrow {}^7F_2$ and ${}^5D_1 \rightarrow {}^7F_1$ emissions of Eu³⁺ in GdVO₄:Eu³⁺ (10 mol %) are 0.44 ms and 10.7 μ s, respectively. The difference is due to different reactions with the lattice of the excited states 5D_0 and 5D_1 of Eu³⁺.
- 3. The infrared spectrum of GdVO₄:Eu³⁺ consists of two peaks at 843.54 and 451.45 cm⁻¹ and indicates that an Eu³⁺ in ⁵D₁ state can relax to ⁵D₀ state by emitting two phonons quickly.

References

- P.A. Studenikin, A.I. Zagumennyi, Yu.D. Zavartsev, P.A. Popov, I.A. Shcherbakov, Quantum Electron (UK) 25 (1995) 1162.
- [2] A.I. Zagumennyi, T.D. Zavartsev, P.A. Studenikin, I.A. Scherbakov, F. Umyskov, Proc. SPIE. Int. Soc. Opt. Eng. (USA) 2698 (1996) 182.
- [3] V.A. Mikhailov, Yu.D. Zavartsev, A.I. Zagumennyi, Quantum Electron (UK) 27 (1997) 13.

- [4] E. Antic-Fidancev, M. Lemaitre-Blaise, P. Porcher, Spectrochim. Acta (Part A) 54A (1998) 2151.
- [5] A. Bril, W.L. Wanmaker, J. Bross, J. Chem. Phys. 43 (1965) 311.
- [6] L.H. Brixner, E. Abramson, J. Electrochem. Soc. 112 (1965) 70.
- [7] F.C. Palilla, A.K. Levin, M. Rinkevics, J. Electrochem. Soc. 112 (1965) 776.
- [8] H. Fuess, A. Kallel, J. Solid State Chem. 5 (1972) 11.
- [9] The Joint Committee on Powder Diffraction Standards, Powder Diffraction File Sets 16 to 18, No. 17–260, Philadelphia, 1974, p. 353.
- [10] S. Qiang, in: Chemistry of Rare Earth, Press of Science and Technology of Henan, Henan, 1993, p. 305, in Chinese.
- [11] C. Linares, A. Louat, M. Blandchard, Struct. Bonding 33 (1977) 179.
- [12] C. Brecher, H. Samelson, A. Lempicki, The energy level structure of Eu³⁺ in YVO₄, in: H.M. Crosswhite, H.W. Moos (Eds.), Optical Properties of Ions in Crystals, Interscience Publishers, New York, 1967, p. 73.
- [13] G.H. Dieke, L.A. Hall, J. Chem. Phys. 27 (1957) 465.
- [14] M.J. Weber, Phys. Rev. B 8 (1973) 54.